

14. RESPIRATION IN PLANTS

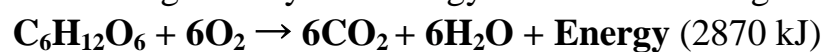
- **Respiration** is the oxidation of nutrients in the cells to release energy and the trapping of this energy for ATP synthesis which is used in biological work.
- The compounds that are oxidized called **respiratory substrates**.
E.g. Carbohydrates (most common), proteins, fats and organic acids

BREATHING IN PLANTS

- For respiration, plants get O₂ and give out CO₂
- In plants, gaseous exchange occurs through **stomata** and **lenticels**.
- Plants need no specialized respiratory organs because-
 - (i) **Each plant part takes care of its own gas – exchange needs.** During photosynthesis, O₂ is released within the cell.
 - (ii) **Plants have only low demand for gas exchange.** They respire at rates much lower than animals do.
 - (iii) **Loose packing of parenchyma** cells in leaves, stems and roots provide an interconnected network of airspace.
 - (iv) **Almost all living cells in a plant have their surfaces exposed to air.**
The interior cells are dead (only for mechanical support).

RESPIRATION IN PLANTS

- The complete combustion of glucose yields energy most of which is given out as heat.

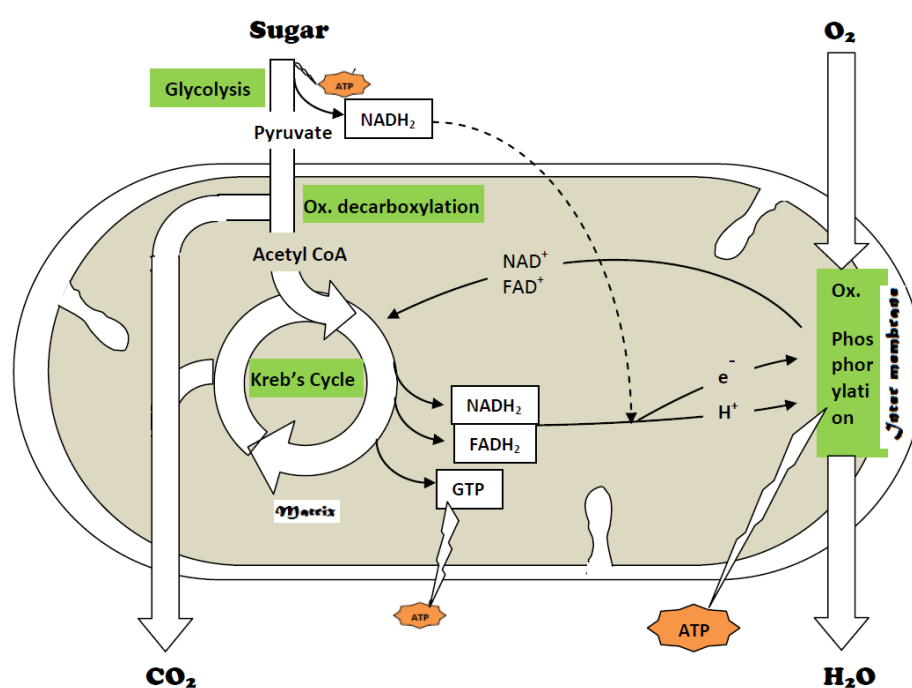


Eukaryotic (plant) cell oxidises glucose in several small steps. It enables some steps to couple released energy to ATP synthesis.

The first cells on earth lived in atmosphere had no O₂. Those and certain present-day living organisms are adapted to anaerobic conditions. Some of them are **facultative** anaerobes. Others are **obligate**.

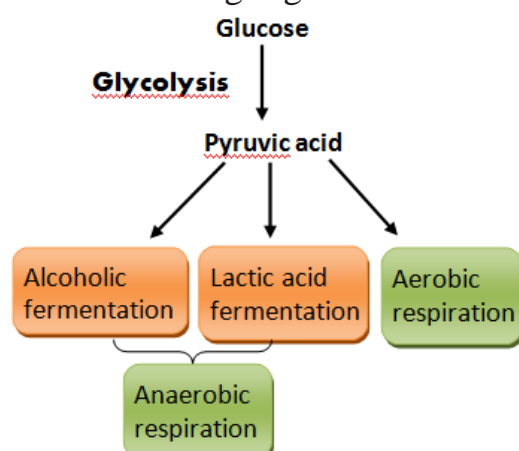
In eukaryotic cells, aerobic (using O₂) respiration is completed in 4 stages-

- 1st Stage** Glycolysis
- 2nd Stage** Oxidative decarboxylation
- 3rd Stage** Krebs' cycle
- 4th Stage** Oxidative phosphorylation



(1st-Stage) GLYCOLYSIS (*Glycos*^G= sugar, *lysis*^G= splitting)

- It is the **partial oxidation** (breakdown) of **glucose** into 2 molecule of **pyruvic acid** in the absence of O₂.
- It **occurs in cytoplasm** of all living organisms. In anaerobes, it is the only process in respiration.



- Scheme was given by Gustav Embden, Otto Meyerhof, and J. Parnas (also known as **EMP pathway**).
- In plants, glucose is derived from **sucrose** (end product of photosynthesis) or from **storage carbohydrate**.
Sucrose is converted into glucose & fructose by the enzyme, **invertase**. These 2 monosaccharides enter the glycolytic pathway.

Mechanism of glycolysis:-

- It includes 10 steps under the control of different enzymes.

(Step-1) **Glucose & fructose** are phosphorylated to form **glucose-6-phosphate** by the enzyme *hexokinase*. An ATP supplies the phosphate.

(Step-2) Glucose-6-phosphate is isomerised into **fructose-6-phosphate**.

(Step-3) Fructose 6-phosphate is phosphorylated to **fructose 1,6-bisphosphate**. An ATP supplies the phosphate.

(Step-4) Fructose 1, 6-bisphosphate is split into 3- carbon **dihydroxyacetone phosphate** and **3-phosphoglyceraldehyde** (PGAL).

(Step-5) Dihydroxyacetone phosphate is then isomerised to **PGAL** itself. Thus 2 PGAL is formed from a single Fructose 1, 6-bisphosphate.

(Step-6) Each of 2 PGAL is oxidised and with inorganic phosphate get converted to **1, 3-bisphosphoglycerate (DPGA)**.

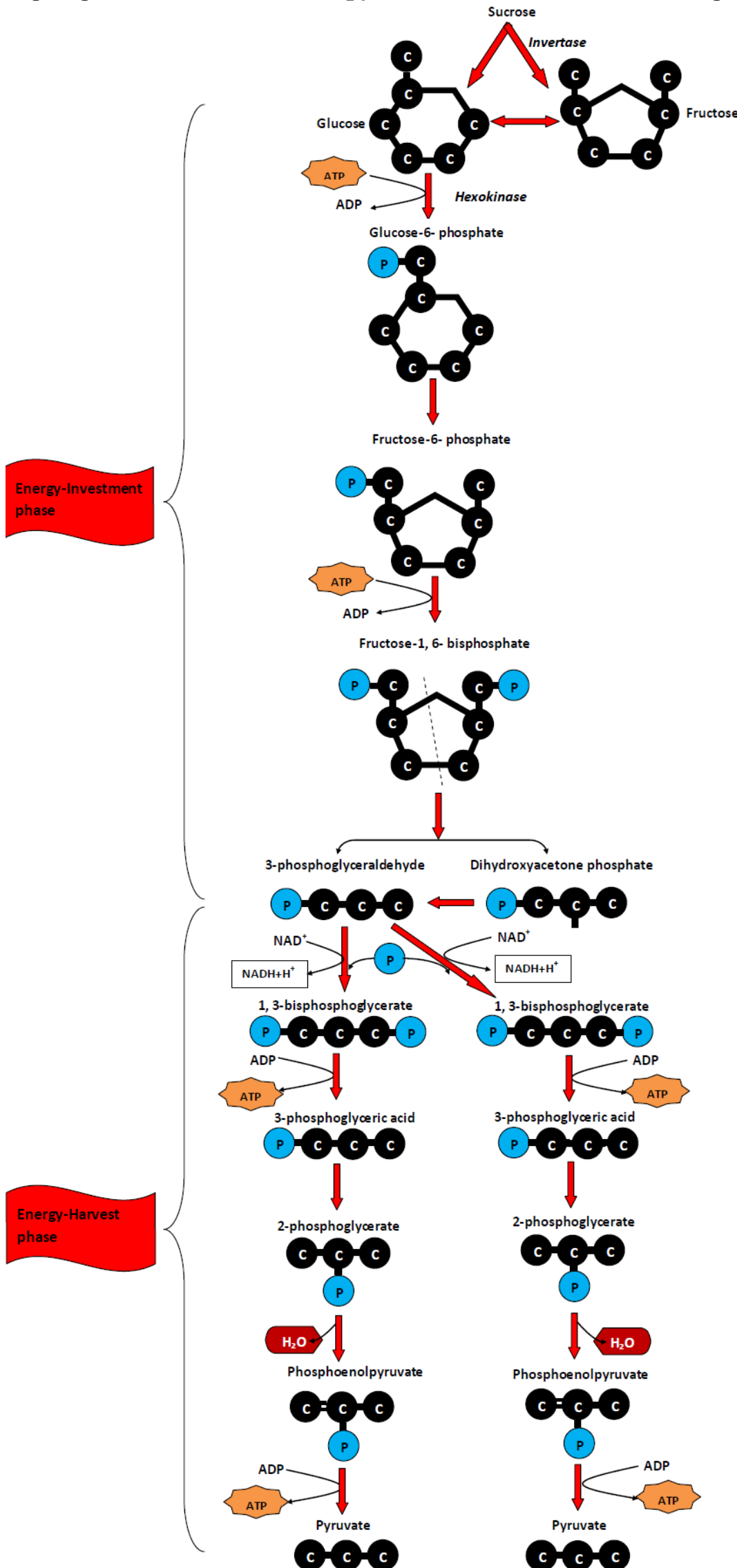
During this, 2 H-atoms are removed from each PGAL and transferred to NAD^+ forming $\text{NADH} + \text{H}^+$.

(Step-7) DPGA loses a phosphate in 1st carbon and is converted to **3-phosphoglyceric acid** (PGA). The loosed P_i is accepted by an ADP to form an ATP.

(Step-8) 3-PGA's phosphate transferred from 3rd C to 2nd C forming **2-PGA**.

(Step-9) 2-PGA losses a **water** molecule to form **Phosphoenol pyruvate** (PEP).

(Step-10) PEP losses a phosphate and is converted to **pyruvic acid**. The loosed P_i is accepted by an ADP to form an ATP.



❖ In glycolysis **2 ATP** is utilised while **4 ATP & 2 $\text{NADH} + \text{H}^+$** are synthesised from 1 glucose molecule.

- In eukaryotic cell, **pyruvic acid (pyruvate)** is the utilised for aerobic respiration.

There is 2 other fate for pyruvic acid under anaerobic condition-

- Alcoholic fermentation
- Lactic acid fermentation

A) **Alcoholic fermentation:** Here, the pyruvic acid formed from glycolysis is converted to CO_2 and ethanol.

The enzymes, **pyruvic acid decarboxylase** and **alcohol dehydrogenase** catalyse these reactions.

E.g. Yeast. Yeasts poison themselves to death when the concentration of alcohol reaches about 13%.

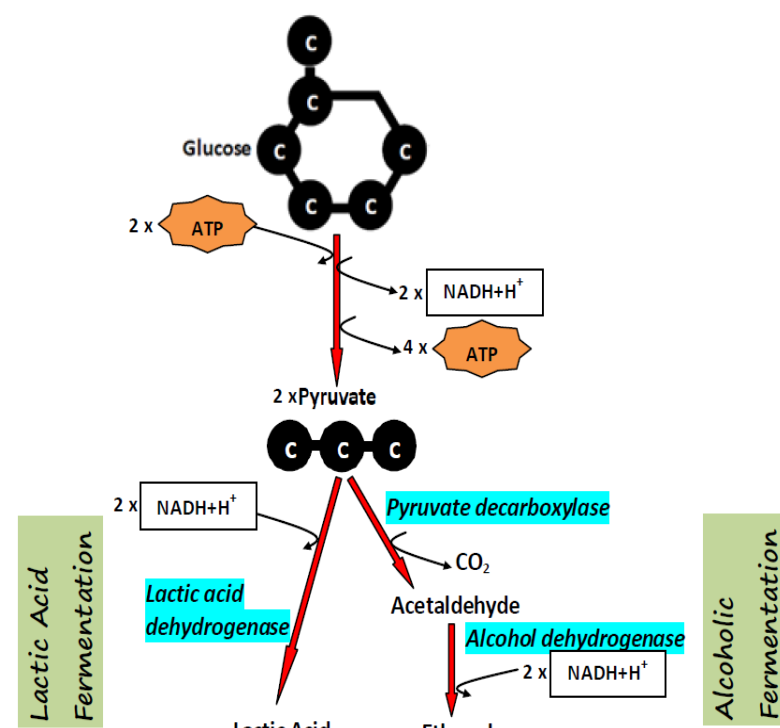
B) **Lactic acid fermentation:** Here, pyruvic acid is converted to lactic acid by **lactate dehydrogenase**.

E.g. Some bacteria, In muscle cells (when O_2 is inadequate during exercise).

Net ATP production from fermentation of one glucose molecule = 2. (4 ATP from glycolysis – 2 ATP utilized).

Drawbacks of fermentation

- Energy production is limited. Less than 7% (56 KCal) of the energy in glucose is released and not all of it is trapped as high energy bonds of ATP.
- Hazardous products (acid or alcohol) are formed.



(2nd-Stage) OXIDATIVE DECARBOXYLATION

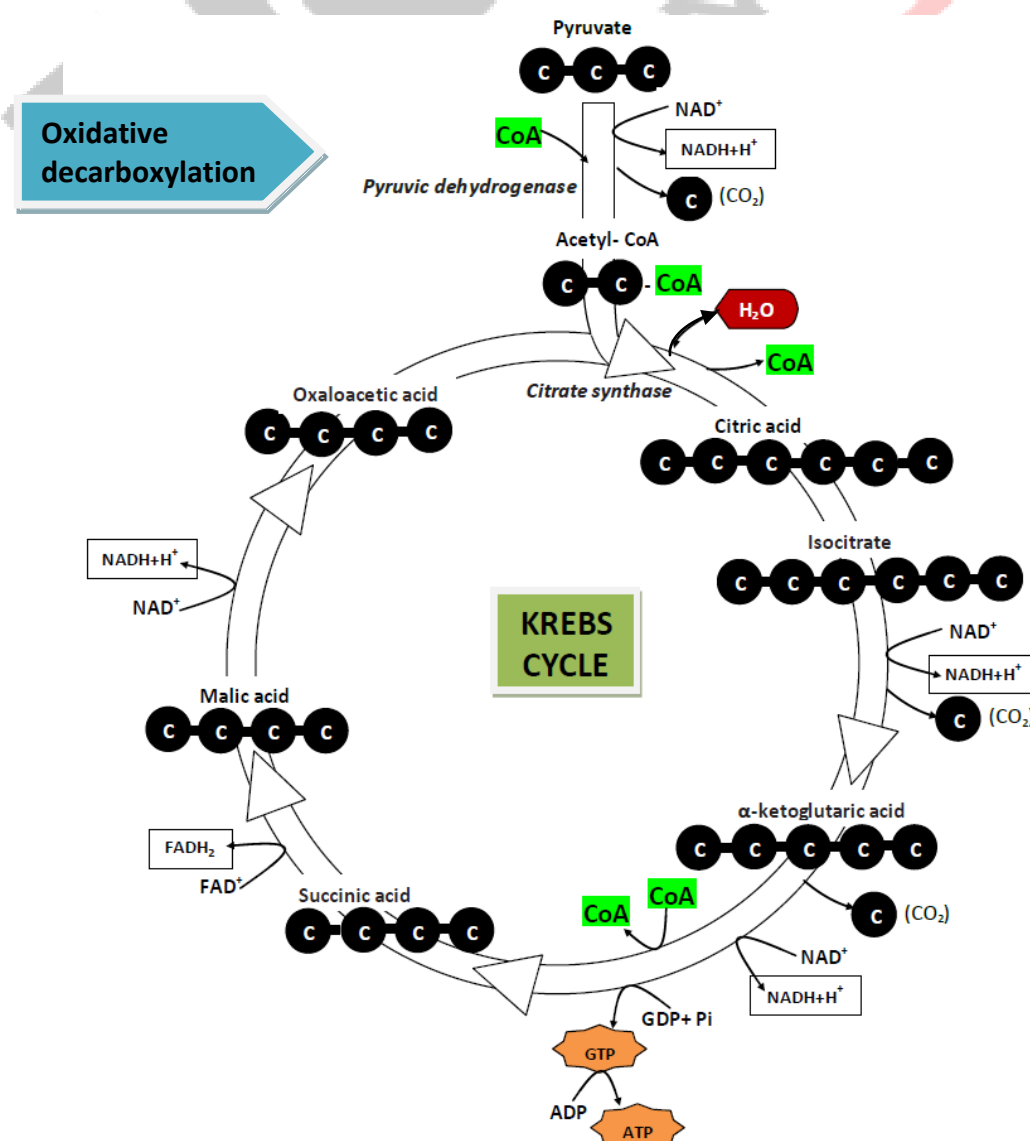
- For this, the **pyruvate** (final product of glycolysis) is transported from the cytoplasm into the mitochondria matrix.
- Pyruvate undergoes **oxidative decarboxylation** to form **acetyl CoA** in presence of **pyruvic dehydrogenase**. It requires coenzymes NAD^+ and **Coenzyme A**.
- During this process, 2 molecules of $\text{NADH}+\text{H}^+$ are produced from 2 molecules of pyruvic acid.
- The acetyl CoA then enters **tricarboxylic acid cycle**.

(3rd-Stage) TRICARBOXYLIC ACID CYCLE (Krebs' cycle)

- TCA cycle was first elucidated by Hans Krebs. (Nobel prize in 1953)

Mechanism of TCA cycle:-

- Acetyl CoA** (2C) condensate with **oxaloacetic acid** (OAA- 4C) and water to form **citric acid** (6C) in presence of the enzyme **citrate synthase**. During this, a molecule of **CoA** is released.
- Citrate** is isomerised to **isocitrate**.
- Isocitrate is decarboxylated to form α -**ketoglutaric acid** (5C). This step reduces NAD^+ to $\text{NADH}+\text{H}^+$.
- α -ketoglutaric acid is decarboxylated to form **succinyl-CoA** (4C). This step also reduce NAD^+ to $\text{NADH}+\text{H}^+$
- Succinyl-CoA is converted to **succinic acid**. This step leads to a substrate level phosphorylation of GDP to form GTP. This reaction is simultaneously coupled with the synthesis of ATP from ADP.
- Succinic acid convert to **malic acid** reduces FAD^+ to FADH_2 .
- Malic acid oxidised to **OAA** and the cycle continues. This step also reduce NAD^+ to $\text{NADH}+\text{H}^+$

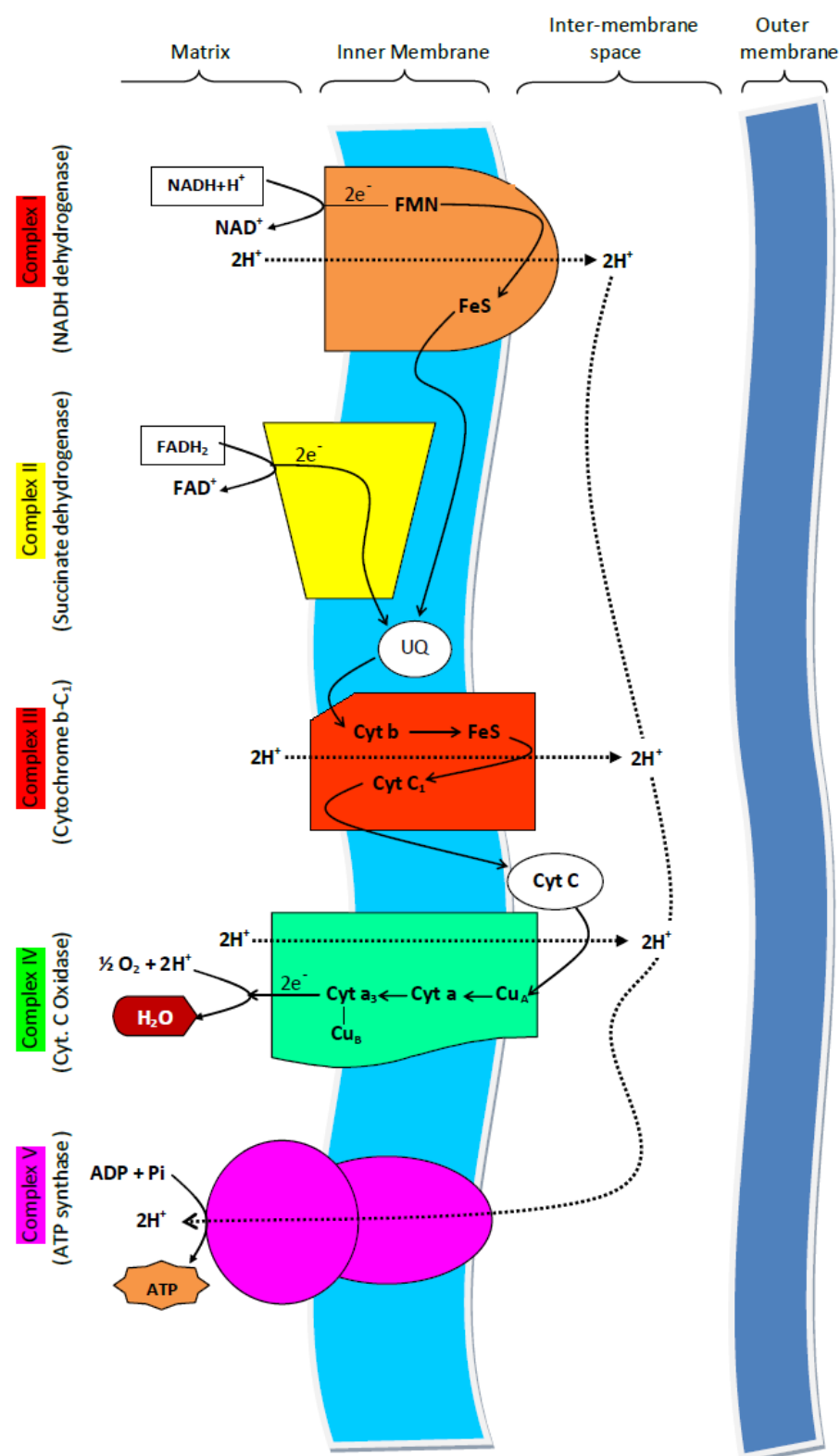


(4th-Stage) OXIDATIVE PHOSPHORYLATION

- This stage is to release and utilize the energy stored in $\text{NADH} + \text{H}^+$ and FADH_2 (formed during Glycolysis & Krebs' cycle) in synthesising ATP. **Electron Transport System (ETS)** facilitate this event.
- ETS** is the metabolic pathway present in the **inner mitochondrial membrane** through which e^- passes from one carrier to another.
 - When the e^- (supplied by $\text{NADH} + \text{H}^+$ and FADH_2) passes from one carrier to another via complex I to IV, the **redox** reaction of carriers progressively extract the energy the e^- possesses and used it to pump H^+ across the mitochondrial membrane. The H^+ gradient is used by ATP synthase (**complex V**) to produce ATP (phosphorylation). So this process is called **oxidative phosphorylation**.
 - The number of ATP molecules synthesised depends on the nature of the e^- donor.
 - ♥ Oxidation of **1 NADH** molecule produces **3 ATP** molecules.
 - ♥ Oxidation of **1 FADH_2** molecule produces **2 ATP** molecules.

Mechanism of Oxidative phosphorylation:-

- (Step-1) e^- from $\text{NADH} + \text{H}^+$ are oxidised by an **NADH dehydrogenase (complex I)**-which contain FMN-Flavin Mono Nucleotide, a protein and is associated with Iron-sulphur protein). e^- are then transferred to **ubiquinone (UQ)** located within the inner membrane.
- (Step-2) Ubiquinone also receives e^- from **FADH_2** via **complex II**.
- (Step-3) The **reduced ubiquinone (ubiquinol)** is then oxidised with the transfer of e^- to **cytochrome bc_1 in complex III** and then to **cytochrome c**.
Cytochrome c is a small protein attached to the outer surface of the inner membrane. It acts as a mobile carrier for transfer of e^- between complex III and IV.
- (Step-4) **Cytochrome c oxidase (Complex IV)** –complex containing **cytochromes a & a_3** , and 2 Cu centres) pass e^- on to O_2 , which is reduced to H_2O by the addition of H^+ . This is called **terminal oxidation**.
- (Step-5) The passage of H^+ from inter membrane space to matrix through the **ATP synthase** is coupled for ATP production. For each ATP produced, 2H^+ passes through **ATP synthase** from the inter-membrane space to the matrix down the electrochemical proton gradient.
- ATP synthase** complex consists of 2 major components, F_1 and F_0 .
 - ✓ F_0 is an integral membrane protein complex that forms the channel through which H^+ cross the inner membrane.
 - ✓ F_1 headpiece is a **peripheral membrane protein complex**. It contains the site for ATP synthesis from ADP & inorganic phosphate.



THE RESPIRATORY BALANCE SHEET

- It is the calculations of the net gain of ATP for every glucose molecule oxidised.
- The net gain of ATP from each glucose molecule can be calculated based on the following assumptions:
 - All steps in Glycolysis, TCA cycle and ETS pathway occur in a sequential and orderly manner
 - The NADH synthesised in glycolysis is transferred into mitochondria and undergoes oxidative phosphorylation.
 - None of the intermediates in the pathway are utilised to synthesise any other compound.
 - Only glucose is being respired – no other alternative substrates are entering in the pathway at any of the intermediary stages.
- Such assumptions are not valid because,
 - All pathways work simultaneously and do not take place one after another.
 - Substrates enter the pathways and are withdrawn from it as and when necessary.
 - ATP is utilized as and when needed.
 - Enzymatic rates are controlled by multiple means.

Such calculations are useful to appreciate the efficiency of the living system in extraction and storing energy.

Net gain of ATP molecules from 1 glucose molecule

Glycolysis:	2 ATP directly	= 2 ATP +
	2 molecules of NADH	= 6 ATP +
Oxidative decarboxylation:	2 NADH	= 6 ATP +
TCA cycle:	6 NADH	= 18 ATP +
	2 FADH	= 4 ATP +
	2 GTP	= 2 ATP
<hr/>		
Total		= 38 ATP -
Transp. of NADH ₂		= 2 ATP ←
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Net gain		= 36 ATP

2 ATP molecules are spent for transporting 2 NADH molecules formed during glycolysis to the mitochondria.

The complete oxidation of 1 molecule of glucose gives = 2870 kJ energy

Total energy stored by 1 ATP = 34 kJ

Total energy stored by 38 ATP = 1292 kJ

$$\text{Efficiency of aerobic respiration} = \frac{1292}{2870} \times 100 = 45\%$$

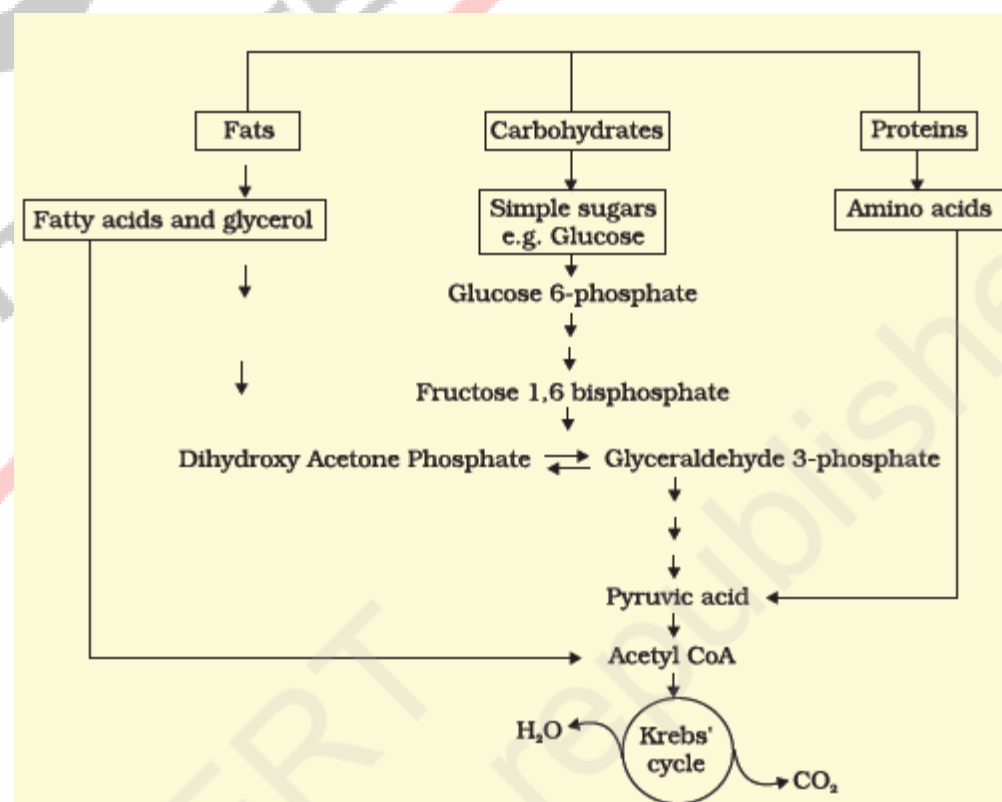
The rest of energy generated during aerobic respiration is lost in the form of heat.

Comparison b/w anaerobic (fermentation) & aerobic respiration

Fermentation	Aerobic respiration
Occurs in the absence of O ₂	Occurs in the presence of O ₂
Partial breakdown of glucose to CO ₂ and ethyl alcohol or lactic acid .	Complete breakdown of glucose to CO ₂ and H ₂ O.
It occurs in the cytoplasm only	It occurs partly in the cytoplasm and the rest in mitochondria
Net gain of only 2 ATP molecules.	Net gain of 36 ATP molecules.
NADH is oxidised to NAD ⁺ rather slowly.	NADH is oxidised to NAD ⁺ very vigorously.

AMPHIBOLIC PATHWAY

- Glucose is the favoured substrate for respiration. All carbohydrates are usually first converted into glucose before they are used for respiration. Other substrates are also respired-
 - Fats** breakdown into glycerol & fatty acid. Fatty acid is degraded to acetyl CoA and enters the pathway. Glycerol is converted to PGAL and enters the pathway.
 - Proteins** are degraded by proteases into amino acids. Each amino acid (after deamination) enters the pathway as pyruvate.
- The respiratory pathway is an **amphibolic pathway** as it involves both anabolism (synthesis) and catabolism (break down).



Respiratory quotient

- It is the ratio of the volume of CO₂ evolved to the volume of O₂ consumed in respiration.

$$RQ = \frac{\text{Volume of CO}_2 \text{ evolved}}{\text{Volume of O}_2 \text{ consumed}}$$

- RQ depends upon the type of respiratory substrate.
 - RQ for carbohydrates = 1**, because equal amounts of CO₂ and O₂ are evolved and consumed, respectively
 - RQ for fats < 1**.
 - RQ for proteins = 0.9**.