Calculate the energy associated with the first orbit of He⁺. What is the radius of this orbit?

ANSWER

(a) calculate the energy associated with the first orbit of $\rm He^+$

$$\begin{split} \mathrm{E} &= -13.6 \frac{\mathrm{z}^2}{\mathrm{n}^2} \\ &= -13.6 \times \frac{2^2}{\mathrm{1}^2} = 54.4 \ \mathrm{ev} \\ \mathrm{Radius \ of \ orbit \ r} &= 529 \times \frac{\mathrm{n}^2}{\mathrm{z}} \ \mathrm{nm} \\ &= 529 \times \frac{\mathrm{12}}{\mathrm{2}} \\ &= 264.5 \ \mathrm{nm} \end{split}$$