

PAPER-1: PHYSICS, MATHEMATICS & CHEMISTRY

Code:

SOLUTIONS TO AIEEE - 2011

Time: 3 hours

Maximum Marks: 360

IMPORTANT INSTRUCTIONS:

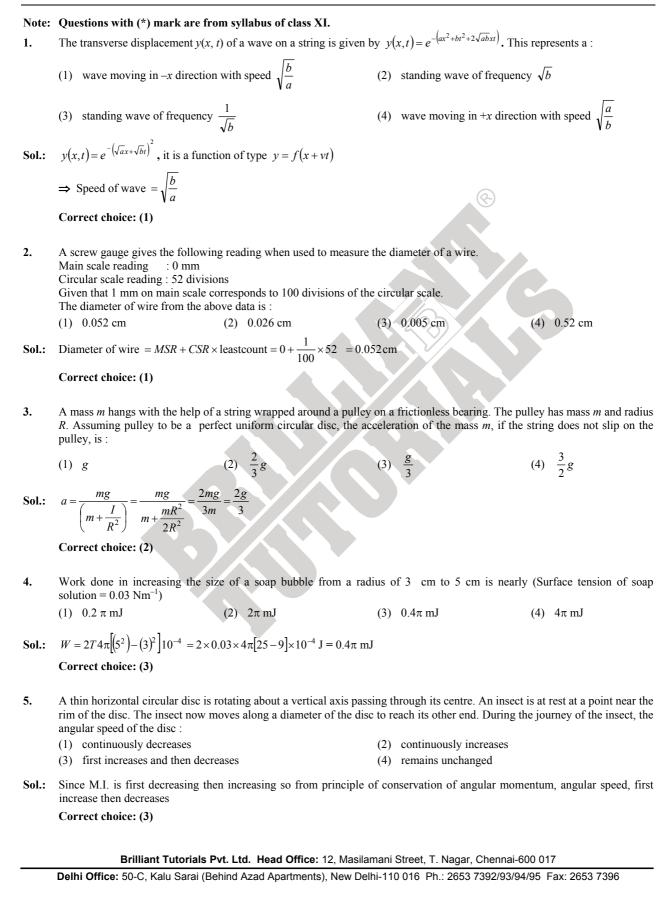
- Immediately fill in the particulars on this page of the Test Booklet with Blue/Black Ball Point Pen. Use of pencil is strictly prohibited.
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- **2.** The Answer Sheet is kept inside this Test Booklet. When you are directed to open the Test Booklet, take out the Answer Sheet and fill in the particulars carefully.
- **3.** The test is of **3 hours** duration.
- **4.** The Test Booklet consists of **90** questions. The maximum marks are **360**.
- 5. There are **three** parts in the questions paper A, B, C consisting of **Physics**, **Mathematics** and **Chemistry** having 30 questions in each part of equal weightage. Each question is allotted **4(four)** marks for each correct response.
- **6.** Candidates will be awarded marks as stated above in instruction No. 5 for correct response of each question. ¹/₄ (one fourth) marks will be deducted for indicating incorrect response of each question. No deduction from the total score will be made if no response is indicated for an item in the answer sheet.
- **7.** There is only one correct response for each question. Filling up more than one response in each question will be treated as wrong response and marks for wrong response will be deducted accordingly as per instruction 6 above.
- Use Blue/Black Ball Point Pen only for writing particulars / marking response on side-1 and Side 2 of the Answer Sheet. Use of pencil is strictly prohibited.
- **9.** No candidate is allowed to carry any textual material printed or written, bits of papers, pager, mobile phone, any electronic device etc; except the Admit Card inside the examination hall/room.
- **10.** Rough work is to be done on the space provided for this purpose in the Test Booklet only. This space is given at the bottom of each page and in 3 pages at the end of the booklet.
- **11.** On completion of the test, the candidate must hand over the Answer Sheet to the invigilator on duty in the Room/Hall. However, the candidates are allowed to take away this Test Booklet with them.
- **12.** The CODE for this Booklet is Q. Make sure that the CODE printed on side 2 of the Answer Sheet is the same as that on this booklet. In case of discrepancy the candidate should immediately report the matter to the invigilator for replacement of both the Test Booklet and the Answer Sheet.
- 13. Do not fold make any stray marks on the Answer Sheet.

Name of the Candidate (in Ca	pital letters):
Roll Number: in figures : in words	
·	
Examination Centre Number	
Name of Examination Centre	(in Capital letters):
Candidate's Signature:	Invigilator's Signature:

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SOLUTIONS TO AIEEE 2011 PHYSICS: (CODE: Q)

PART – A



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6. Two particles are executing simple harmonic motion of the same amplitude A and frequency ω along the x-axis. Their mean position is separated by distance $X_0(X_0 > A)$. If the maximum separation between them is $(X_0 + A)$, the phase difference between their motion is :

(1)
$$\frac{\pi}{3}$$
 (2) $\frac{\pi}{4}$ (3) $\frac{\pi}{6}$ (4) $\frac{\pi}{2}$

Sol.: Let, $x_1 = A \sin \omega t$ and $x_2 = A \sin(\omega t + \phi)$

$$x_{2} - x_{1} = 2A\cos\left(\omega t + \frac{\phi}{2}\right)\sin\frac{\phi}{2}$$

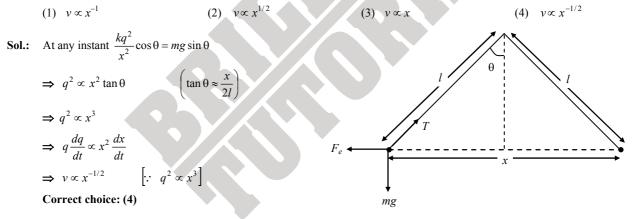
But $2A\sin\frac{\phi}{2} = A \implies \phi = \frac{\pi}{3}$

Correct choice: (1)

7. Two bodies of masses m and 4m are placed at a distance r. The gravitational potential at a point on the line joining them where the gravitational field is zero is :

(1)
$$-\frac{4Gm}{r}$$
 (2) $-\frac{6Gm}{r}$ (3) $-\frac{9Gm}{r}$ (4) zero
Sol.: $\frac{Gm}{x^2} = \frac{4Gm}{(r-x)^2} \Rightarrow \frac{1}{x} = \frac{2}{r-x}$ \therefore $r-x=2x$
 $x = \frac{r}{3}$
Gravitational potential $V = -\frac{Gm}{\frac{r}{3}} - \frac{4Gm}{\frac{2r}{3}} = -\frac{9Gm}{r}$
Correct choice: (3)

8. Two identical charged spheres suspended from a common point by two massless strings of length l are initially a distance $d(d \ll l)$ apart because of their mutual repulsion. The charge begins to leak from both the spheres at a constant rate. As a result charges approach each other with a velocity v. Then as a function of distance x between them,



A boat is moving due east in a region where the earth's magnetic field is $5.0 \times 10^{-5} \text{ NA}^{-1} \text{ m}^{-1}$ due north and horizontal. The boat carries a vertical aerial 2 m long. If the speed of the boat is 1.50 ms^{-1} , the magnitude of the induced emf in the wire of 9. aerial is :

$$(1) \quad 0.75 \text{ mV} \qquad (2) \quad 0.50 \text{ mV} \qquad (3) \quad 0.15 \text{ mV} \qquad (4) \quad 1 \text{ mV}$$

emf = $vB_H l = 1.5 \times 5 \times 10^{-5} \times 2 = 15 \times 10^{-5} = 0.15 \text{ mV}$ Sol.:

Correct choice: (3)

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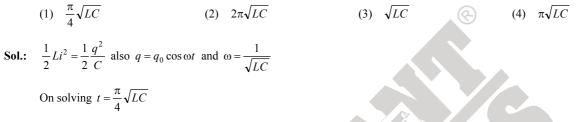
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10. An object, moving with a speed of 6.25 m/s, is decelerated at a rate given by :

 $\frac{dv}{dt} = -2.5\sqrt{v} \text{ where } v \text{ is the instantaneous speed. The time taken by the object, to come to rest, would be :}$ (1) 2 s
(2) 4 s
(3) 8 s
(4) 1 s
Sol.: $\frac{dv}{dt} = -2.5\sqrt{v}$ $\int_{625}^{0} v^{-1/2} dv = -2.5 \int_{0}^{t} dt$ t = 2 secCorrect choice: (1)

11. A fully charged capacitor C with initial charge q_0 is connected to a coil of self inductance L at t = 0. The time at which the energy is stored equally between the electric and the magnetic fields is :



Correct choice: (1)

12. Let the *x-z* plane be the boundary between two transparent media. Medium 1 in $z \ge 0$ has a refractive index of $\sqrt{2}$ and medium 2 with z < 0 has a refractive index of $\sqrt{3}$. A ray of light in medium 1 given by the vector $\vec{A} = 6\sqrt{3}\hat{i} + 8\sqrt{3}\hat{j} - 10\hat{k}$ is incident on the plane of separation. The angle of refraction in medium 2 is : (1) 45° (2) 60° (3) 75° (4) 30°

 $\mu = \sqrt{2}$

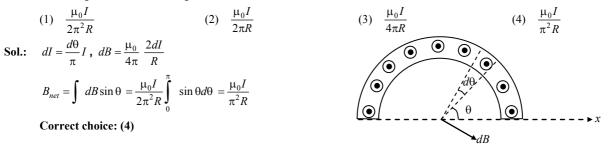
 $\mu = \sqrt{3}$

Sol.: Angle of incidence is given by
$$\left(c \sqrt{2}\hat{i} + 8\sqrt{2}\hat{j} - 10\hat{k}\right)\hat{k}$$

 $\cos (\pi - i) = \frac{\left[\frac{6\sqrt{3}i + 8\sqrt{3}j - 10k \right]k}{20} \right]$ $-\cos i = -\frac{1}{2}$ $\angle i = 60^{0}$ From Snell's law, $\sqrt{2} \sin i = \sqrt{3} \sin r$ $\angle r = 45^{0}$

Correct choice: (1)

13. A current *I* flows in an infinitely long wire with cross section in the form of a semi-circular ring of radius *R*. The magnitude of the magnetic induction along its axis is :



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- 5
- 14. A thermally insulated vessel contains an ideal gas of molecular mass M and ratio of specific heats γ . It is moving with speed ν and its suddenly brought to rest. Assuming no heat is lost to the surroundings, its temperature increases by :

(1)
$$\frac{(\gamma - 1)}{2\gamma R} M v^2 K$$
 (2) $\frac{\gamma M v^2}{2R} K$ (3) $\frac{(\gamma - 1)}{2R} M v^2 K$ (4) $\frac{(\gamma - 1)}{2(\gamma + 1)R} M v^2 K$

Sol.: Work done is zero

So loss in kinetic energy = change in internal energy of gas

$$\frac{1}{2}mv^{2} = nC_{v}\Delta T = n\frac{R}{\gamma-1}\Delta T$$
$$\frac{1}{2}mv^{2} = \frac{m}{M}\frac{R}{\gamma-1}\Delta T$$
$$\therefore \quad \Delta T = \frac{Mv^{2}(\gamma-1)}{2R}K$$

Correct choice: (3)

15. A mass *M*, attached to a horizontal spring, executes S.H.M. with amplitude A_1 . When the mass *M* passes through its mean position then a smaller mass *m* is placed over it and both of them move together with amplitude A_2 . The ratio of $\left(\frac{A_1}{A_1}\right)$ is :

(1)
$$\frac{M+m}{M}$$
 (2) $\left(\frac{M}{M+m}\right)^{1/2}$ (3) $\left(\frac{M+m}{M}\right)^{1/2}$ (4) $\frac{M}{M+m}$

Sol.: At mean point $F_{\text{net}} = 0$, so linear momentum must be conserved

$$Mv_{1} = (M + m)v_{2}$$

$$MA_{1}\sqrt{\frac{K}{M}} = (M + m)A_{2}\sqrt{\frac{K}{m + M}}$$

$$A_{1}\sqrt{M} = A_{2}\sqrt{M + m}$$

$$\therefore \quad \frac{A_{1}}{A_{2}} = \sqrt{\frac{m + M}{M}}$$

- Correct choice: (3)
- 16. Water is flowing continuously from a tap having an internal diameter 8×10^{-3} m. The water velocity as it leaves the tap is 0.4 ms^{-1} . The diameter of the water stream at a distance 2×10^{-1} m below the tap is close to :

(1)
$$7.5 \times 10^{-3}$$
 m (2) 9.6×10^{-3} m (3) 3.6×10^{-3} m (4) 5.0×10^{-3} m

Sol.: From Bernoulli's theorem

$$P_0 + \frac{1}{2}\rho v_1^2 + \rho gh = P_0 + \frac{1}{2}\rho v_2^2 + 0$$
$$v_2 = \sqrt{v_1^2 + 2gh} = \sqrt{0.16 + 2 \times 10 \times 0.2} = 2.03 \text{ m/s}$$

From continuity equation $A_2v_2 = A_1v_1$

$$\pi \frac{D_2^2}{4} \times v_2 = \pi \frac{D_1^2}{4} v_1$$

$$\Rightarrow D_2 = D_1 \sqrt{\frac{v_1}{v_2}} = 3.55 \times 10^{-3} \text{ m}$$

Correct choice: (3)

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This question has Statement – 1 and Statement – 2. Of the four choices given after the statements, choose the one that best describes the two statements.
 Statement – 1 :

Sky wave signals are used for long distance radio communication. These signals are in general, less stable than ground wave signals.

Statement – 2 :

- The state of ionosphere varies from hour to hour, day to day and season to season.
- (1) Statement -1 is true, Statement -2 is true, Statement -2 is the correct explanation of Statement -1.
- (2) Statement -1 is true, Statement -2 is true, Statement -2 is not the correct explanation of Statement -1.
- (3) Statement -1 is false, Statement -2 is true.
- (4) Statement -1 is true, Statement -2 is false.

Sol.: Correct choice: (2)

18. Three perfect gases at absolute temperatures T_1 , T_2 and T_3 are mixed. The masses of molecules are m_1 , m_2 and m_3 and the number of molecules are n_1 , n_2 and n_3 respectively. Assuming no loss of energy, the final temperature of the mixture is :

(1)
$$\frac{n_1T_1 + n_2T_2 + n_3T_3}{n_1 + n_2 + n_3}$$
 (2) $\frac{n_1T_1^2 + n_2T_2^2 + n_3T_3^2}{n_1T_1 + n_2T_2 + n_3T_3}$ (3) $\frac{n_1^2T_1^2 + n_2^2T_2^2 + n_3^2T_3^2}{n_1T_1 + n_2T_2 + n_3T_3}$ (4) $\frac{(T_1 + T_2 + T_3)}{3}$
Sol.: If atomicity of gas is same
So $\frac{f}{2}(n_1 + n_2 + n_3)kT = \frac{f}{2}n_1kT_1 + \frac{f}{2}n_2kT_2 + \frac{f}{2}n_3kT_3$
 $\therefore T = \frac{n_1T_1 + n_2T_2 + n_3T_3}{n_1 + n_2 + n_3}$
Correct choice: (1)

- 19. A pulley of radius 2 m is rotated about its axis by a force $F = (20 t 5t^2)$ Newton (where t is measured in seconds) applied tangentially. It the moment of inertia of the pulley about its axis of rotation is 10 kg-m² the number of rotations made by the pulley before its direction of motion if reserved, is:
 - (1) more than 3 but less than 6
 - (3) more than 9

(2) more than 6 but less than 9

(4) less than 3

Sol.: $F = 20t - 5t^2$

$$\therefore \quad \alpha = \frac{FR}{I} = 4t - t^2 \qquad \Rightarrow \frac{d\omega}{dt} = 4t - t^2 \qquad \Rightarrow \quad \int_0^{\omega} d\omega = \int_0^t (4t - t^2) dt$$
$$\Rightarrow \quad \omega = 2t^2 - \frac{t^3}{3} \qquad (as \ \omega = 0 \ at \ t = 0, \ 6s)$$
$$\int_0^{\theta} d\theta = \int_0^6 \left(2t^2 - \frac{t^3}{3} \right) dt \qquad \Rightarrow \theta = 36 \ rad \qquad \Rightarrow n = \frac{36}{2\pi} < 6$$
Correct choice: (1)

- 20. A resistor '*R*' and $2\mu F$ capacitor in series in connected through a switch to 200 V direct supply. Across the capacitor is a neon bulb that lights up at 120 V. Calculate the value of *R* to make the bulb light up 5 s after the switch has been closed. $(\log_{10} 2.5 = 0.4)$
- (1) $1.7 \times 10^5 \Omega$ (2) $2.7 \times 10^6 \Omega$ (3) $3.3 \times 10^7 \Omega$ (4) $1.3 \times 10^4 \Omega$ Sol.: $V_c = V_{\max} \left(1 - e^{-t/RC} \right);$ $120 = 200 \left(1 - e^{-t/RC} \right) \Rightarrow t = RC \ln(2.5) \Rightarrow R = 2.71 \times 10^6 \Omega$

Correct choice: (2)

- 21. A Carnot engine operating between temperatures T_1 and T_2 has efficiency $\frac{1}{6}$. When T_2 is lowered by 62 K its efficiency increases to $\frac{1}{3}$. Then T_1 and T_2 are, respectively:
 - (1) 372 K and 330 K (2) 330 K and 268 K (3) 310 K and 248 K (4) 372 K and 310 K

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7

Sol.: Efficiency of engine $\eta = 1 - \frac{T_2}{T_1}$

$$\Rightarrow \frac{T_2}{T_1} = \frac{5}{6} \qquad \dots (i)$$

$$\eta_2 = 1 - \frac{T_2 - 62}{T_1} = \frac{1}{2} \qquad \dots (ii)$$

 $==\frac{1}{3}$... (II)

Solving we get,

 $T_1 = 372 \,\mathrm{K}$ and $T_2 = \frac{5}{6} \times 372 = 310 \,\mathrm{K}$

Correct choice: (4)

22. If a wire is stretched to make it 0.1% longer, its resistance will :
(1) increase by 0.2%
(2) decrease by 0.2%

(3) decrease by 0.05%

(4) increase by 0.05%

Sol.: Resistance of wire $R = \frac{\rho l}{A} = \frac{\rho l^2}{C}$ (volume constant or Al = C)

:. Fractional change in resistance $\frac{\Delta R}{R} = 2\frac{\Delta l}{l}$

Correct choice: (1)

23. This question has a paragraph followed by two statements, Statement -1 and Statement -2. Of the given four alternatives after the statements, choose the one that describes the statements.

A thin air film is formed by putting the convex surface of a plane-convex lens over a plane glass plate. With monochromatic light, this film gives an interference pattern due to light reflected from the top (convex) surface and the bottom (glass plate) surface of the film.

Statement – 1 : When light reflects from the air-glass plate interface, the reflected wave suffers a phase change of π .

Statement – 2 : The centre of the interference pattern is dark.

- (1) Statement -1 is true, Statement -2 is true, Statement -2 is the correct explanation of Statement -1.
- (2) Statement -1 is true, Statement -2 is true, Statement -2 is not the correct explanation of Statement -1.
- (3) Statement -1 is false, Statement -2 is true.
- (4) Statement -1 is true, Statement -2 is false.
- Sol.: Correct choice: (2)
- 24. A car is fitted with a convex side-view mirror of focal length 20 cm. A second car 2.8 m behind the first car is overtaking the first car at a relative speed of 15 m/s. The speed of the image of the second car as seen in the mirror of the first one is :
 - (1) $\frac{1}{15}$ m/s (2) 10 m/s (3) 15 m/s (4) $\frac{1}{10}$ m/s
- **Sol.:** From mirror formulae

$$\frac{1}{v} + \frac{1}{u} = \frac{1}{f} \text{ so, } \frac{dv}{dt} = -\frac{v^2}{u^2} \left(\frac{du}{dt}\right) \Rightarrow \frac{dv}{dt} = -\left(\frac{f}{u-f}\right)^2 \frac{du}{dt} \Rightarrow \frac{dv}{dt} = \frac{1}{15} \text{ m/s}$$

Correct choice: (1)

25. Energy required for the electron excitation in Li^{++} from the first to the third Bohr orbit is : (1) 36.3 eV (2) 108.8 eV (3) 122.4 eV (4) 12.1 eV Sol.: $\Delta E = 13.6(3)^2 \left(\frac{1}{1^2} - \frac{1}{3^2}\right) = 108.8 \text{ eV}$

Correct choice: (2)

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8

(4) $\pi \frac{v^2}{g}$

(4) 7 min

- 26. The electrostatic potential inside a charged spherical ball is given by $\phi = ar^2 + b$ where *r* is the distance from the centre *a*, *b* are constants. Then the charge density inside the ball is :
- (1) $-6a\varepsilon_0 r$ (2) $-24\pi a\varepsilon_0$ (3) $-6a\varepsilon_0$ (4) $-24\pi a\varepsilon_0 r$ Sol.: Electric field $E = -\frac{d\phi}{dr} = -2ar$... (i) By Gauss's theorem $E = \frac{1}{4\pi\varepsilon_0} \frac{q}{r^2}$... (ii) From (i) and (ii) $q = -8\pi\varepsilon_0 ar^3$, $dq = -24\pi\varepsilon_0 ar^2 dr$ Charge density $\rho = \frac{dq}{4\pi r^2 dr} = -6\varepsilon_0 a$ Correct choice: (3)
- 27. A water fountain on the ground sprinkles water all around it. If the speed of water coming out of the fountain is v, the total area around the fountain that gets wet is :

1)
$$\pi \frac{v^4}{g^2}$$

Sol.: Total area around fountain $A = \pi R_{\text{max}}^2$

Where
$$R_{\text{max}} = \frac{v^2 \sin 90^0}{g} = \frac{v^2}{g}$$
 \therefore $A = \pi \frac{v^4}{g^2}$

(2) $\frac{\pi}{2} \frac{v^4}{g^2}$

(2) 20 min

Correct choice: (1)

- 28. 100g of water is heated from 30°C to 50°C. Ignoring the slight expansion of the water, the change in its internal energy is (specific heat of water is 4184 J/kg/K):
 (1) 8.4 kJ
 (2) 84 kJ
 (3) 2.1 kJ
 (4) 4.2 kJ
- (1) 8.4 kJ Sol.: $\Delta U = \Delta Q = mc\Delta T = 100 \times 10^{-3} \times 4184(50 - 30) \approx 8.4$ kJ Correct choice: (1) (2) 84 kJ (3) 2.1 kJ (4) 4.2 kJ
- 29. The half life of a radioactive substance is 20 minutes. The approximate time interval $(t_2 t_1)$ between the time t_2 when $\frac{2}{3}$ of it

 $\frac{N_0}{2} = N_0 e^{-\lambda t_2}$

28 min

has decayed and time t_1 when $\frac{1}{3}$ of it had decayed is :

Sol.: No of undecayed atom after time t_2 ;

Number of undecayed atom after time t_1 ;

Solving (i) and (ii) $t_2 - t_1 = 20 \min$

Correct choice: (2)

30. This question has Statement -1 and Statement -2. Of the four choices given after the statements, choose the one that best describes the two statements.

Statement – 1:

A metallic surface is irradiated by a monochromatic light of frequency $v > v_0$ (the threshold frequency). The maximum kinetic energy and the stopping potential are K_{max} and V_0 respectively. If the frequency incident on the surface is doubled, both the K_{max} and V_0 are also doubled.

Statement – 2 :

The maximum kinetic energy and the stopping potential of photoelectrons emitted from a surface are linearly dependent on the frequency of incident light.

- (1) Statement -1 is true, Statement -2 is true, Statement -2 is the correct explanation of Statement -1.
- (2) Statement -1 is true, Statement -2 is true, Statement -2 is not the correct explanation of Statement -1.
- (3) Statement -1 is false, Statement -2 is true.
- (4) Statement -1 is true, Statement -2 is false.

Sol.: $K_{\max} = eV_0 = hv - hv_0$

When v is doubled, K_{max} and V_0 become more that double.

Correct choice: (3)

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PART – B

Note: Questions with (*) mark are from syllabus of class XI. *31. The lines $L_1: y - x = 0$ and $L_2: 2x + y = 0$ intersect the line $L_3: y + 2 = 0$ at P and Q respectively. The bisector of the acute angle between L_1 and L_2 intersects L_3 at R. **Statement-1:** The ratio PR : RQ equals $2\sqrt{2}$: $\sqrt{5}$ Statement-2: In any triangle, bisector of an angle divides the triangle into two similar triangles. (1) Statement-1 is true, Statement-2 is true; Statement-2 is not a correct explanation for Statement-1. (2) Statement-1 is true, Statement-2 is false. (3) Statement-1 is false, Statement-2 is true. (4) Statement-1 is true, Statement-2 is true; Statement-2 is a correct explanation for Statement-1. **Sol.:** P = (-2, -2)Q = (1, -2)A(0, 0) $\frac{PR}{RQ} = \frac{AP}{AQ} = \frac{2\sqrt{2}}{\sqrt{5}}$ But statement 2 is not always true 2xCorrect choice: (2) *32. If $A = \sin^2 x + \cos^4 x$, then for all real x : (4) $\frac{3}{4} \le A \le 1$ (1) $\frac{13}{16} \le A \le 1$ (3) $\frac{3}{4} \le A \le \frac{13}{16}$ $(2) \quad 1 \le A \le 2$ **Sol.:** $A = (1 - \sin^2 x)^2 + \sin^2 x$ $=\sin^4 x - \sin^2 x + 1$ $=\left(\sin^2 x - \frac{1}{2}\right)^2 + \frac{3}{4}$ $0 \leq \sin^2 x \leq 1$ $0 \le \left(\sin^2 x - \frac{1}{2}\right)^2 \le \frac{1}{4}$ $\frac{3}{4} \le A \le 1$. Correct choice: (4) *33. The coefficient of x^7 in the expansion of $(1 - x - x^2 + x^3)^6$ is (1) -132(2) -144(3) 132 (4) 144 **Sol.:** $(1-x-x^2+x^3)^6 = (1-x)^6(1-x^2)^6$ $= \left(\sum_{r=0}^{6} {}^{6}C_{r} (-x)^{r}\right) \left(1 - 6x^{2} + 15x^{4} - 20x^{6} + ...\right)$ $= -6 \times {}^{6}C_{5} \times (-1)^{5} + 15 \times {}^{6}C_{3} \times (-1)^{3} - 20 \times {}^{6}C_{1} \times (-1)^{1}$ = 36 - 300 + 120 = -144Correct choice: (2)

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34.
$$\lim_{x \to 2} \left(\frac{\sqrt{1 - \cos\{2(x-2)\}}}{x-2} \right)$$

(1) equals $\sqrt{2}$

(2) equals $-\sqrt{2}$

(3) equals $\frac{1}{\sqrt{2}}$

(4) does not exist

(4) $\left(\frac{d^2 y}{dx^2}\right)^{-1}$

Sol.:
$$\lim_{x \to 2} \sqrt{\frac{1 - \cos 2(x - 2)}{x - 2}} = \lim_{x \to 2} \frac{\sqrt{2} |\sin(x - 2)|}{x - 2}$$

LHL = $-\sqrt{2}$

RHL = $\sqrt{2}$

Correct choice: (4)

*35. Statement-1: The number of ways of distributing 10 identical balls in 4 distinct boxes such that no box is empty is ${}^{9}C_{3}$.

Statement-2: The number of ways of choosing any 3 places from 9 different places is ${}^{9}C_{3}$.

- (1) Statement-1 is true, Statement-2 is true; Statement-2 is not a correct explanation for Statement-1.
- (2) Statement-1 is true, Statement-2 is false.
- (3) Statement-1 is false, Statement-2 is true.
- (4) Statement-1 is true, Statement-2 is true; Statement-2 is a correct explanation for Statement-1.

Sol.:
$$x_1 + x_2 + x_3 + x_4 = 10, x_i \ge 1, i = 1, 2, 3, 4$$

Number of ways ${}^{10-1}C_{4-1} = {}^{9}C_{3}$

This can also be explained as : Between 10 balls, there are 9 places out of which 3 are to be selected to make 4 partitions. **Correct choice: (4)**

(3)

36.
$$\frac{d^2x}{dy^2}$$
 equals :

(1)
$$-\left(\frac{d^2 y}{dx^2}\right)^{-1}\left(\frac{dy}{dx}\right)$$

Sol.:
$$\frac{d^2x}{dy^2} = \frac{d}{dy} \left(\frac{dx}{dy} \right) = \frac{d}{dx} \left(\frac{dx}{dy} \right) \frac{dx}{dy}$$

$$=\frac{d}{dx}\left(\frac{1}{dy/dx}\right)\frac{dx}{dy}$$

$$= -\frac{1}{\left(\frac{dy}{dx}\right)^3} \frac{d^2 y}{dx^2}$$

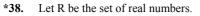
Correct choice: (3)

37. If
$$\frac{dy}{dx} = y + 3 > 0$$
 and $y(0) = 2$, then $y(\ln 2)$ is equal to :
(1) 5 (2) 13 (3) -2 (4) 7
Sol.: $\frac{dy}{dx} = y + 3$
 $\int \frac{dy}{y+3} = \int dx \implies \ln(y+3) = x + c$
 $\ln 5 = 0 + c$
 $\therefore \ln(y+3) = x + \ln 5$
 $\ln(y+3) = \ln 2 + \ln 5 \implies y + 3 = 10 \implies y = 7$
Correct choice: (4)

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Statement-1: $A = \{(x, y) \in R \times R : y - x \text{ is an integer}\}$ is an equivalence relation on *R*.

Statement-2: $B = \{(x, y) \in R \times R : x = \alpha y \text{ for some rational number } \alpha\}$ is an equivalence relation on *R*.

(1) Statement-1 is true, Statement-2 is true; Statement-2 is not a correct explanation for Statement-1.

- (2) Statement-1 is true, Statement-2 is false.
- (3) Statement-1 is false, Statement-2 is true.

(4) Statement-1 is true, Statement-2 is true; Statement-2 is a correct explanation for Statement-1.

Sol.: Let for statement 1 $xRy = x - y \in I$. As xRx is an integer and yRx as well as xRz (for xRy and yRz) is also an integer Hence equivalence.

(3) log 2

(4) $\pi \log 2$

Similarly as $x = \alpha y$ hence $\alpha = 1$ for reflexive and product of rationals also being rational \Rightarrow transitive

But not symmetric because of $\alpha = 0$ case. e.g. for (0, 1), $\alpha = 0$ but no α for (1, 0).

Correct choice: (2)

39. The value of
$$\int_{0}^{1} \frac{8 \log(1+x)}{1+x^2} dx$$
 is
(1) $\frac{\pi}{8} \log 2$ (2) $\frac{\pi}{2} \log 2$

 $I = 8 \int_{-\infty}^{1} \frac{\log(1+x)}{dx} dx$

Sol.:

Put
$$x = \tan \theta$$

 $I = 8 \int_{0}^{\pi/4} \log(1 + \tan \theta) d\theta$
 $= 8 \times \frac{\pi}{8} \log 2 = \pi \log 2$

Correct choice: (4)

- *40. Let α, β be real and z be a complex number. If $z^2 + \alpha z + \beta = 0$ has two distinct roots on the line Re z = 1, then it is necessary that :
 - (1) $\beta \in (-1, 0)$ (2) $|\beta| = 1$ (3) $\beta \in (1, \infty)$ (4) $\beta \in (0, 1)$

Sol.: $z^2 + \alpha z + \beta = 0$

Since the coefficients are real, complex roots must occur in conjugate pairs. Let the roots be 1 + ix, 1 - ix, $x \neq 0$. $\beta = (1 + ix)(1 - ix) = 1 + x^2 > 1$

Correct choice: (3)

41. Consider 5 independent Bernoulli's trials each with probability of success p. If the probability of at least one failure is greater than or equal to $\frac{31}{32}$, then p lies in the interval

(1)
$$\left(\frac{3}{4}, \frac{11}{12}\right)$$

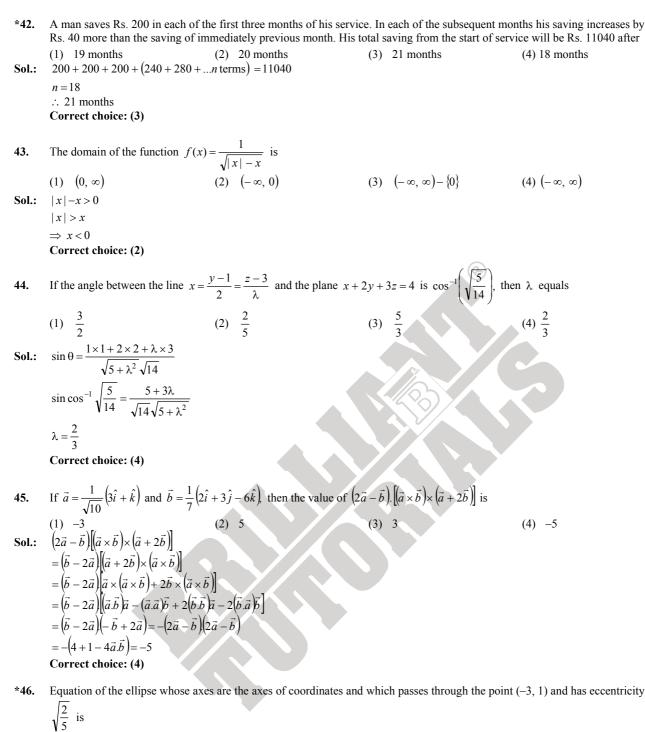
Sol.: $P(X=r) = {}^{5}C_{r}p^{r}q^{5-r}$
 $1 - P(X=5) \ge \frac{31}{32} \implies p^{5} \le \frac{1}{32}$
 $p \le \frac{1}{2}$
Correct choice: (2)
(3) $\left(\frac{11}{12}, 1\right)$
(4) $\left(\frac{1}{2}, \frac{3}{4}\right)$

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12



(1) $5x^2 + 3y^2 - 48 = 0$ (2) $3x^2 + 5y^2 - 15 = 0$ (3) $5x^2 + 3y^2 - 32 = 0$ (4) $3x^2 + 5y^2 - 32 = 0$ Sol.: $\frac{x^2}{a^2} + \frac{y^2}{b^2} = 1$ As $b^2 = a^2 \left(1 - \frac{2}{5}\right)$...(i) $\frac{9}{a^2} + \frac{1}{b^2} = 1$...(ii) $\frac{9}{a^2} + \frac{5}{a^2(3)} = 1$

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(4) $T^2 - \frac{I}{k}$

$$\frac{1}{a^2}\left(9+\frac{5}{3}\right) = 1 \implies a^2 = \frac{32}{3}$$
$$b^2 = \frac{32}{3} \times \frac{3}{5} = \frac{32}{5}$$
$$\implies 3x^2 + 5y^2 - 32 = 0.$$

There is one more case of vertical ellipse which gives $5x^2 + 3y^2 - 48 = 0$. So two options are correct. Correct choice: (1) and (4)

47. Let *I* be the purchase value of an equipment and V(t) be the value after it has been used for *t* years. The value V(t) depreciates at a rate given by differential equation $\frac{dV(t)}{dt} = -k(T-t)$, where k > 0 is a constant and *T* is the total life in years of the equipment. Then the scrap value V(T) of the equipment is

(3) e^{-kT}

(1)
$$I - \frac{kT^2}{2}$$

Sol.: $\frac{dV(t)}{dt} + kT - kt = 0$
 $\frac{dV(t)}{dt} - kt = -kT = 0$

 $\frac{dV(t)}{dt} - kt = -kT \Rightarrow LDE \quad (\text{at } t = 0, V = I)$ $\Rightarrow V(T) = I - \frac{kT^2}{2} \qquad (\text{scrap value obtained by putting } t = T)$

(2) $I - \frac{k(T-t)^2}{2}$

Correct choice: (1)

48. The vectors \vec{a} and \vec{b} are not perpendicular and \vec{c} and \vec{d} are two vectors satisfying $\vec{b} \times \vec{c} = \vec{b} \times \vec{d}$ and $\vec{a} \cdot \vec{d} = 0$. Then the vector \vec{d} is equal to

(1)
$$\vec{c} + \left(\frac{\vec{a}.\vec{c}}{\vec{a}.\vec{b}}\right)\vec{b}$$
 (2) $\vec{b} + \left(\frac{\vec{b}.\vec{c}}{\vec{a}.\vec{b}}\right)\vec{c}$ (3) $\vec{c} - \left(\frac{\vec{a}.\vec{c}}{\vec{a}.\vec{b}}\right)\vec{b}$ (4) $\vec{b} - \left(\frac{\vec{b}.\vec{c}}{\vec{a}.\vec{b}}\right)\vec{c}$

Sol.: $\vec{a}.\vec{b} \neq 0$

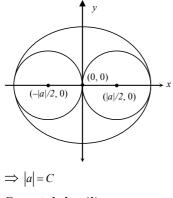
 $\vec{b} \times \vec{c} = \vec{b} \times \vec{d} \text{ and } \vec{a}.\vec{d} = 0$ $\Rightarrow \vec{a} \times (\vec{b} \times \vec{c}) = \vec{a} \times (\vec{b} \times \vec{d})$ $\Rightarrow (\vec{a}.\vec{c})\vec{b} - (\vec{a}.\vec{b})\vec{c} = (\vec{a}.\vec{d})\vec{b} - (\vec{a}.\vec{b})\vec{d}$ $\Rightarrow (\vec{a}.\vec{b})\vec{d} = -(\vec{a}.\vec{c})\vec{b} + (\vec{a}.\vec{b})\vec{c}$ $\vec{d} = \vec{c} - \left(\frac{\vec{a}.\vec{c}}{\vec{a}.\vec{b}}\right)\vec{b}$

Correct choice: (3)

*49. The two circles $x^2 + y^2 = ax$ and $x^2 + y^2 = c^2(c > 0)$ touch each other if

(1)
$$|a| = c$$
 (2) $a = 2c$ (3) $|a| = 2c$ (4) $2|a| = c$

Sol.:



Correct choice: (1)

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- 14
- **50.** If C and D are two events such that $C \subset D$ and $P(D) \neq 0$, then the correct statement among the following is

(1)
$$P(C|D) \ge P(C)$$
 (2) $P(C|D) < P(C)$ (3) $P(C|D) = \frac{P(D)}{P(C)}$ (4) $P(C|D) = P(C)$
Sol: In this case
 $P\left(\frac{C}{D}\right) = \frac{P(C \cap D)}{P(D)} = \frac{P(C)}{P(D)}$
where $P(D) \le 1$ hence $P\left(\frac{C}{D}\right) \ge P(C)$
Correct choice: (1)
51. The number of values of k for which the linear equations $4x + ky + 2z = 0$, $kx + 4y + z = 0$ and $2x + 2y + z = 0$ possess a non-zero solution is
(1) 2 (2) 1 (3) zero (4) 3
Sol: $A = 0$
 $\Rightarrow \frac{1}{8} \frac{k}{2} \frac{2}{1} = 0 \Rightarrow 4(4 - 2) - k(k - 2) + 2(2k - 8) = 0$
 $\Rightarrow \frac{3}{8} - k^2 + 2k + 4k - 16 = 0$
 $k^2 - 6k + 8 = 0 \Rightarrow (4 - 2) - k(k - 2) + 2(2k - 8) = 0$
 $\Rightarrow \frac{3}{8} - k^2 + 2k + 4k - 16 = 0$
 $k^2 - 6k + 8 = 0 \Rightarrow (4 - 2) - k(k - 2) + 2(2k - 8) = 0$
 $\Rightarrow \frac{3}{8} - k^2 + 2k + 4k - 16 = 0$
 $k^2 - 6k + 8 = 0 \Rightarrow (4 - 2) - k(k - 2) + 2(2k - 8) = 0$
 $\Rightarrow \frac{3}{8} - k^2 + 2k + 4k - 16 = 0$
 $k^2 - 6k + 8 = 0 \Rightarrow (4 - 2) - k(k - 2) + 2(2k - 8) = 0$
 $\Rightarrow \frac{3}{8} - k^2 + 2k + 4k - 16 = 0$
 $k^2 - 6k + 8 = 0 \Rightarrow (4 - 2) - k(k - 2) + 2(2k - 8) = 0$
 $\Rightarrow \frac{3}{8} - k^2 + 2k + 4k - 16 = 0$
 $k^2 - 6k + 8 = 0 \Rightarrow (4 - 2) - k(k - 2) + 2(2k - 8) = 0$
 $\Rightarrow \frac{3}{8} - k^2 + 2k + 4k - 16 = 0$
 $k^2 - 6k + 8 = 0 \Rightarrow (4 - 2) - k(k - 2) + 2(2k - 8) = 0$
 $\Rightarrow \frac{3}{8} - k^2 + 2k + 4k - 16 = 0$
 $k^2 - 6k + 8 = 0 \Rightarrow (4 - 2) - k(k - 2) + 2(2k - 8) = 0$
 $\Rightarrow \frac{3}{8} - k^2 + 2k + 4k - 16 = 0$
 $k^2 - 6k + 8 = 0 \Rightarrow (k - 2)(k - 2) = 0, k = 4, 2$
Correct choice: (1)
*53. The shortest distance between line $p - x = 1$ and gave $x = y^2$ is
(1) $\frac{3\sqrt{2}}{8}$ (2) $\frac{8}{3\sqrt{2}}$ (3) $\frac{4}{\sqrt{3}}$ (4) $\frac{\sqrt{3}}{4}$
Sol: $y = mx - 2mn - am^2$ is $m = -1$
 $y = -x - 2\frac{1}{4}(-1)\frac{1}{4}(-1)^2$
 $y = -x + \frac{1}{2} + \frac{1}{4} = \frac{3}{4} - x$
Common normal
 $y = \frac{3}{4} - x \Rightarrow AB \Rightarrow A = \left(-\frac{1}{8}, \frac{7}{8}\right); B = \left(\frac{1}{4}, \frac{1}{2}\right\right)$
 $= \frac{3}{8}\sqrt{2}$
Correct choice: (1)
*54. If the mean deviation about the median of the numbers $a, 2a, \dots, 50a$ is 50, then $|a|$ equals
(1) 3 (2) 4 (3) 5 (4) 2
Sol: Median $(M) = \frac{25a + 25a}{2} = \frac{5n}{2}$ and Now $MD = \sum \frac{|x_1 - M|}{n}$

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Statement-1: The point A(1, 0, 7) is the mirror image of the point B(1, 6, 3) in the line : $\frac{x}{1} = \frac{y-1}{2} = \frac{z-2}{3}$. 55. Statement-2: The line $\frac{x}{1} = \frac{y-1}{2} = \frac{z-2}{3}$ bisects the line segment joining A(1, 0, 7) and B(1, 6, 3). (1) Statement-1 is true, Statement-2 is true; Statement-2 is not a correct explanation for Statement-1. (2) Statement-1 is true, Statement-2 is false. (3) Statement-1 is false, Statement-2 is true. (4) Statement-1 is true, Statement-2 is true; Statement-2 is a correct explanation for Statement-1. Sol.: As the line $\frac{x}{1} = \frac{y-1}{2} = \frac{z-2}{2}$ is the right bisector of line segment joining (1, 0, 7) with (1, 6, 3). But right bisector is not mentioned in statement 2. Correct choice: (1) Let A and B be two symmetric matrices of order 3. 56. **Statement-1:** A(BA) and (AB)A are symmetric matrices. Statement-2: AB is symmetric matrix if matrix multiplication of A with B is commutative. (1) Statement-1 is true, Statement-2 is true; Statement-2 is not a correct explanation for Statement-1. (2) Statement-1 is true, Statement-2 is false. (3) Statement-1 is false, Statement-2 is true. (4) Statement-1 is true, Statement-2 is true; Statement-2 is a correct explanation for Statement-1. **Sol.:** $(A(BA))^T = (BA)^T A^T = (A^T B^T) A^T = (AB) A = A(BA)$ Correct choice: (1) *57. If $\omega(\neq 1)$ is a cube root of unity, and $(1 + \omega)^7 = A + B\omega$. Then (A, B) equals (3) (-1, 1)(1) (1, 1)(2) (1, 0)(4) (0, 1) **Sol.:** $\Rightarrow (1 + \omega)^7 = A + B\omega$ $\left(-\omega^2\right)^7 = 1 + \omega$ A = 1, B = 1Correct choice: (1) The values of p and q for which the function $f(x) = \begin{cases} \frac{\sin(p+1)x + \sin x}{x} , & x < 0\\ q , & x = 0 \end{cases}$, x = 0 is continuous for all x in R, are $\frac{\sqrt{x+x^2} - \sqrt{x}}{x^{3/2}}$, x > 058. (2) $p = -\frac{3}{2}, q = \frac{1}{2}$ (3) $p = \frac{1}{2}, q = \frac{3}{2}$ (4) $p = \frac{1}{2}, q = -\frac{3}{2}$ (1) $p = \frac{5}{2}, q = \frac{1}{2}$ Sol.: LHL:- $\lim_{x \to 0^{-}} f(x) = \lim_{h \to 0} \frac{\sin\{(p+1)(-h)\} - \sinh}{-h} = p + 1 + 1 = p + 2$ RHL:- $\lim_{x \to 0^{+}} f(x) = \lim_{h \to 0} = \frac{1}{1+1} = \frac{1}{2}$ (by rationalization) $f(0) = q \Longrightarrow p = -\frac{3}{2}, q = \frac{1}{2}$ Correct choice: (2)

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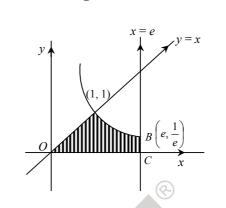
59. The area of the region enclosed by the curves y = x, x = e, $y = \frac{1}{x}$ and the positive x-axis is

(1) 1 square unit (2) $\frac{3}{2}$ square units (3) $\frac{5}{2}$ square units (4) $\frac{1}{2}$ square unit

Sol.: Required region AOBC:-

$$\int_{0}^{1} x dx + \int_{1}^{e} \frac{1}{x} dx$$
$$= \frac{1}{2} + 1 = \frac{3}{2}$$

Correct choice: (2)



60. For
$$x \in \left(0, \frac{5\pi}{2}\right)$$
, define $f(x) = \int_{0}^{x} \sqrt{t} \sin t \, dt$. Then f has

- (1) local minimum at π and 2π
- (2) local minimum at π and local maximum at 2π
- (3) local maximum at π and local minimum at 2π
- (4) local maximum at π and 2π

Sol.:
$$f'(x) = \sqrt{x} \sin x$$

 $f'(x) = 0$

$$\Rightarrow x = 0$$
 or $\sin x = 0$

$$\Rightarrow x = 2\pi, \pi$$

$$f''(x) = \sqrt{x} \cos x + \frac{1}{2\sqrt{x}} \sin x$$

$$\Rightarrow \quad \frac{1}{2\sqrt{x}}(2x\cos x + \sin x)$$

at $x = \pi \Rightarrow f''(x) < 0 \Rightarrow$ maxima

at $x = 2\pi \Rightarrow f''(x) > 0 \Rightarrow$ minima

Correct choice: (3)

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SOLUTIONS TO AIEEE 2011 CHEMISTRY: (CODE: Q) PART - C

	PART – C		
Note:	Questions with (*) mark are from syllabus of class XI.		
*61. Sol.:	Among the following the maximum covalent character is shown b (1) $SnCl_2$ (2) $AlCl_3$ Higher charge density of cation favours covalency. Correct choice: (2)	by the compound: (3) MgCl ₂	(4) FeCl ₂
62.	The presence or absence of hydroxy group on which carbon atom (1) 2^{nd} (2) 3^{rd}	$(3) 4^{th}$	DNA? (4) 1 st
Sol.:	RNA has hydroxy group attached at the 2 nd C-atom while DNA de Correct choice: (1)	bes not.	
63.	Trichloroacetaldehyde was subjected to Cannizzaro's reaction by trichloroacetate and another compound. The other compound is: (1) Trichloromethanol (3) Chloroform Cl Cl Cl Cl	using NaOH. The mixture of th (2) 2,2,2-Trichloropropanol (4) 2,2,2- Trichloroethanol	e products contains sodiun
Sol.:	$\begin{array}{ccc} Cl & Cl & Cl \\ Cl-C-C-H & & \\ I & \\ Cl & Cl & \\ Cl & Cl & \\ Cl & Cl & \\ \end{array} \xrightarrow{\begin{subarray}{c} Cl \\ NaOH \end{subarray}} Cl-C-COONa + Cl-C-CH_2OH \\ Cl & Cl & \\ Cl & Cl \end{subarray}$		
64.	Sodium ethoxide has reacted with ethanoyl chloride. The compou(1) 2-Butanone(2) Ethyl chloride	nd that is produced in the above (3) Ethyl ethanoate	reaction is: (4) Diethyl ether
Sol.:	$\begin{array}{ccc} CH_{3}-C-Cl &+ & C_{2}H_{5} \overset{\bullet}{O} Na^{+} \longrightarrow & CH_{3}-C-OC_{2}H_{5} &+ & NaCl \\ & & & \\ O & & & O \\ \end{array}$ Correct choice: (3)		×
65. Sol.:	The reduction potential of hydrogen half-cell will be negative if: (1) $p(H_2) = 1$ atm and $[H^+] = 1.0$ M (3) $p(H_2) = 2$ atm and $[H^+] = 2.0$ M Pt $H_2(g)$ $H^+(aq)$	(2) p(H ₂) = 2 atm and [H ⁺] = 1 (4) p(H ₂) = 1 atm and [H ⁺] = 2	
501	$H^{+}H_{2}(g) + H^{-}(aq)^{2}$ $H^{+} + e^{-} \longrightarrow \frac{1}{2}H_{2}(g)$ $E = E^{\circ} - \frac{0.059}{1} \log \frac{(P_{H_{2}})^{1/2}}{[H^{+}]} ; E = 0 - \frac{0.059}{1} \log \frac{2^{1/2}}{1} = -ve$		
	Correct choice: (2)		
*66.	The strongest acid amongst the following compounds is:(1) HCOOH(2) CH3CH2CH(Cl)CO2H	(3) CICH ₂ CH ₂ CH ₂ COOH	(4) CH ₃ COOH
Sol.:	$CH_3CH_2CH(Cl)CO_2H$ is the strongest acid due to $-I$ effect of Cl. Correct choice: (2)		
67.	The degree of dissociation (α) of a weak electrolyte, $A_x B_y$ is relate (1) $\alpha = \frac{i-1}{x+y+1}$ (2) $\alpha = \frac{x+y-1}{i-1}$	ed to van't Hoff factor (<i>i</i>) by the (3) $\alpha = \frac{x + y + 1}{i - 1}$	expression: (4) $\alpha = \frac{i-1}{(x+y-1)}$
Sol.:	$\begin{array}{rcl} x+y+1 & & l-1 \\ & & & \\ A_{x}B_{y} & \Longrightarrow & xA^{y+} & + & yB^{x-} \\ \end{array}$ Moles at equilibrium $\begin{array}{rcl} 1-\alpha & x\alpha & y\alpha \\ i=1-\alpha+x\alpha+y\alpha \\ (i-1)=\alpha(x+y-1) \\ \alpha=\frac{i-1}{x+y-1} \\ \end{array}$ Correct choice: (4)	<i>i</i> - 1	(x + y - 1)

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*68.	'a' and 'b' are van der Waal's co	nstants for gases. Chlorine is mo	re easily liquefied than ethane bec	ause
	(1) a and b for $Cl_2 < a$ and b for C (3) a for $Cl_2 > a$ for C_2H_6 but b for		(2) a for $Cl_2 < a$ for C_2H_6 but b f (4) a and b for $Cl_2 > a$ and b for	
Sol.:			e of attraction while 'b' is a meas	
	molecule.			
	Correct choice: (3)			
*69.	A vessel at 1000 K contains CO graphite. If the total pressure at e	D_2 with a pressure of 0.5 atm.	Some of the CO_2 is converted in	to CO on the addition of
	(1) 3 atm	(2) 0.3 atm	(3) 0.18 atm	(4) 1.8 atm
Sol.:	$CO_2(g) + C(s) \rightleftharpoons 2CO(g)$			
	$\begin{array}{ccc} 0.5 - x & 2x \\ 0.5 - x + 2x = 0.8 \end{array}$			
	<i>x</i> = 0.3			
	$K = \frac{0.6 \times 0.6}{0.2} = 1.8$ atm.			
	Correct choice: (4)			
*70.	Boron cannot form which one of	the following anions?		
/0.	(1) BH_{4}^{-}	(2) $B(OH)_4^-$	(3) BO_{2}^{-}	(4) BF_6^{3-}
Sol.:	Boron cannot form BF_6^{3-} due to			
	Correct choice: (4)	Ĩ		
71.	Which of the following facts abo (1) The complex is paramagnetic		wrong?	
	(2) The complex is an outer orbit	al complex.		
	 (3) The complex gives white pred (4) The complex involves d²sp³ h 			
Sol.:	The complex involves d^2sp^3 hybr	-		
	Correct choice: (2)			
72.	Ethylene glycol is used as an ant	ifreeze in a cold climate. Mass o	f ethylene glycol which should be g mol ⁻¹ , and molar mass of ethyler	added to 4 kg of water to
	(1) 204.30 g	(2) 400.00 g	(3) 304.60 g	(4) 804.32 g
Sol.:	$\Delta T_f = K_f \times m$	\mathbf{X}		
	$6 = 1.86 \times \frac{W}{62 \times 4}$			
	$w = \frac{6 \times 62 \times 4}{1.86} = 800 \text{ g.}$			
	1.86 Correct choice: (4)			
*73.		rs presents the correct sequence of	of the increasing basic nature of the	e given oxides?
,	(1) MgO $<$ K ₂ O $<$ Al ₂ O ₃ $<$ Na ₂ O		$(2) \operatorname{Na_2O} < \operatorname{K_2O} < \operatorname{MgO} < \operatorname{Al_2O}$	3
6.1	$(3) K_2O < Na_2O < Al_2O_3 < MgO$	1 * * * * 1 *	(4) $Al_2O_3 < MgO < Na_2O < K_2O$)
Sol.:	The basic character of metal oxic Correct choice: (4)	le increases with increase in elec	tropositivity of metal.	
74.	The rate of a chemical reaction of the reaction increases by about:	doubles for every 10°C rise of to	emperature. If the temperature is r	aised by 50°C, the rate of
. .	(1) 24 times Increase in reaction rate = $2^5 = 32$	(2) 32 times	(3) 64 times	(4) 10 times
Sol.:	Increase in reaction rate = 2^3 = 32 Correct choice: (2)	2 times.		
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19

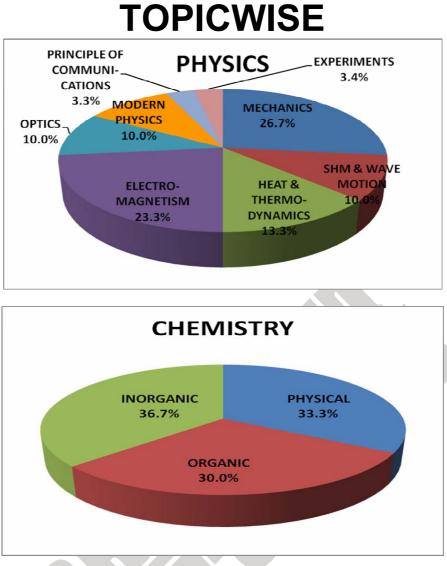
75.	The magnetic moment (spin onl (1) 5.46 BM	y) of [NiCl ₄] ^{2–} is: (2) 2.82 BM	(3) 1.41 BM	(4) 1.82 BM
Sol.:	$Ni^{2+} 3d^84s^0$	(2) 2.82 Divi	(<i>5</i>) 1.41 DIVI	(4) 1.62 DM
	The hybridisation of central me	tal ion is sp ³ in NiCl ₄ ⁻² .		
	Therefore number of unpaired e	electrons = 2 .		
	$\mu = \sqrt{n(n+2)} = \sqrt{2 \times 4} = 2\sqrt{2}$	$z = 2 \times 1.41 = 2.82$ BM.		
	Correct choice: (2)			
*76.	The hybridisation of orbitals of	N atom in NO_3^- , NO_2^+ and	NH ⁺ ₄ are respectively:	
	(1) sp^2 , sp , sp^3	(2) sp, sp^3 , sp^2	$(3) sp^2, sp^3, sp$	(4) sp, sp^2 , sp^3
Sol.:	$-0-N \stackrel{\oplus}{\underset{O^-}{\overset{\oplus}{\overset{\oplus}}}} , 0=N=0$,	$H^{H}_{H^{\oplus}}$		
	Hybridization of N atom in NC	P_3^- , NO ₂ ⁺ and NH ₄ ⁺ are res	spectively sp^2 , sp , sp^3 .	2
	Correct choice: (1)			
77.	In context of the lanthanoids, w (1) All the members exhibit ± 2	•	nents is <i>not</i> correct?	2
	(1) All the members exhibit +3(2) Because of similar propertie		ids is not easy.	
	(3) Availability of 4f electrons i			
Sol.:	(4) There is a gradual decrease			in the series. 2 and +4 oxidation state but they
501	easily revert to the more stable		of the faithanolds also show f	2 and 14 oxidation state but they
	Correct choice: (3)			
*78.	solution?			e fraction of methyl alcohol in the
C 1	(1) 0.190	(2) 0.086	(3) 0.050	(4) 0.100
Sol.:	$n_{CH_3OH} = 5.2$ moles			
	$n_{\rm H_2O} = \frac{1000}{18} = 55.55$ moles			
	$X_{\rm CH_3OH} = \frac{5.2}{5.2 + 55.55} = 0.086$			
70	$X_{CH_{3}OH} = \frac{5.2}{5.2 + 55.55} = 0.086$ Correct choice: (2)			
79.	$X_{CH_3OH} = \frac{5.2}{5.2 + 55.55} = 0.086$ Correct choice: (2) Which of the following statemet	nt is <i>wrong</i> ?		
79.	$X_{CH_{3}OH} = \frac{5.2}{5.2 + 55.55} = 0.086$ Correct choice: (2) Which of the following stateme (1) Nitrogen cannot form $d\pi$ -p π (2) Single N–N bond is weaker	nt is <i>wrong</i> ? bond. than the single P–P bond.		
79.	$X_{CH_{3}OH} = \frac{5.2}{5.2 + 55.55} = 0.086$ Correct choice: (2) Which of the following stateme (1) Nitrogen cannot form $d\pi$ -p π (2) Single N–N bond is weaker (3) N ₂ O ₄ has two resonance stru-	nt is <i>wrong</i> ? bond. than the single P–P bond. ictures.		
	$X_{CH_3OH} = \frac{5.2}{5.2 + 55.55} = 0.086$ Correct choice: (2) Which of the following stateme (1) Nitrogen cannot form $d\pi$ -p π (2) Single N–N bond is weaker (3) N ₂ O ₄ has two resonance stru (4) The stability of hydrides inc	nt is <i>wrong</i> ? bond. than the single P–P bond. ictures. reases from NH ₃ to BiH ₃ in		atom
79. Sol.:	$X_{CH_{3}OH} = \frac{5.2}{5.2 + 55.55} = 0.086$ Correct choice: (2) Which of the following stateme (1) Nitrogen cannot form $d\pi$ -p π (2) Single N–N bond is weaker (3) N ₂ O ₄ has two resonance stru-	nt is <i>wrong</i> ? bond. than the single P–P bond. ictures. reases from NH ₃ to BiH ₃ in		atom.
	$X_{CH_3OH} = \frac{5.2}{5.2 + 55.55} = 0.086$ Correct choice: (2) Which of the following stateme (1) Nitrogen cannot form $d\pi$ -p π (2) Single N–N bond is weaker (3) N ₂ O ₄ has two resonance stru (4) The stability of hydrides inc The stability of hydrides decrea Correct choice: (4) The outer electron configuration	nt is <i>wrong</i> ? bond. than the single P–P bond. tetures. reases from NH ₃ to BiH ₃ in ses from NH ₃ to BiH ₃ due t n of Gd (Atomic number: 64	o increase in size of the central a 4) is:	
Sol.: 80.	$X_{CH_3OH} = \frac{5.2}{5.2 + 55.55} = 0.086$ Correct choice: (2) Which of the following statemet (1) Nitrogen cannot form $d\pi$ -p π (2) Single N–N bond is weaker (3) N ₂ O ₄ has two resonance struct (4) The stability of hydrides income The stability of hydrides decreas Correct choice: (4) The outer electron configuration (1) 4f ⁸ 5d ⁰ 6s ²	nt is <i>wrong</i> ? bond. than the single P–P bond. tetures. reases from NH_3 to BiH_3 in ses from NH_3 to BiH_3 due t n of Gd (Atomic number: 6- (2) $4f^4 5d^4 6s^2$	o increase in size of the central a	atom. $(4) 4f^3 5d^5 6s^2$
Sol.:	$X_{CH_3OH} = \frac{5.2}{5.2 + 55.55} = 0.086$ Correct choice: (2) Which of the following stateme (1) Nitrogen cannot form $d\pi$ -p π (2) Single N–N bond is weaker (3) N ₂ O ₄ has two resonance stru (4) The stability of hydrides inco The stability of hydrides decrea Correct choice: (4) The outer electron configuration (1) 4f ⁸ 5d ⁰ 6s ² Outer electronic configuration of	nt is <i>wrong</i> ? bond. than the single P–P bond. tetures. reases from NH_3 to BiH_3 in ses from NH_3 to BiH_3 due t n of Gd (Atomic number: 6- (2) $4f^4 5d^4 6s^2$	o increase in size of the central a 4) is:	
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Sol.: 80. Sol.:	$X_{CH_3OH} = \frac{5.2}{5.2 + 55.55} = 0.086$ Correct choice: (2) Which of the following statemet (1) Nitrogen cannot form $d\pi$ -p π (2) Single N–N bond is weaker (3) N ₂ O ₄ has two resonance struct (4) The stability of hydrides income The stability of hydrides decreat Correct choice: (4) The outer electron configuration (1) 4f ⁸ 5d ⁰ 6s ² Outer electronic configuration of Correct choice: (3) Which of the following statemet (1) The vapour at 200°C consist	nt is <i>wrong</i> ? bond. than the single P–P bond. ictures. reases from NH ₃ to BiH ₃ in ses from NH ₃ to BiH ₃ due t n of Gd (Atomic number: 6^{-} (2) $4f^4 5d^4 6s^2$ of Gd is $4f^7 5d^1 6s^2$ nts regarding sulphur is <i>inc</i> is mostly of S ₈ rings. nsists of S ₂ molecules. ur is never less than +4 in it	a) is: (3) $4f^7 5d^1 6s^2$ (3) orrect?	
Sol.: 80. Sol.:	$X_{CH_3OH} = \frac{5.2}{5.2 + 55.55} = 0.086$ Correct choice: (2) Which of the following stateme (1) Nitrogen cannot form $d\pi$ -p π (2) Single N–N bond is weaker (3) N ₂ O ₄ has two resonance strue (4) The stability of hydrides income The stability of hydrides decrean Correct choice: (4) The outer electron configuration (1) 4f ⁸ 5d ⁰ 6s ² Outer electronic configuration of Correct choice: (3) Which of the following stateme (1) The vapour at 200°C consist (2) At 600°C the gas mainly con (3) The oxidation state of sulphu (4) S ₂ molecule is paramagnetic The oxidation state of sulphur in	nt is <i>wrong</i> ? bond. than the single P–P bond. actures. reases from NH ₃ to BiH ₃ in ses from NH ₃ to BiH ₃ due t n of Gd (Atomic number: 64^{-1} (2) $4f^4 5d^4 6s^2$ of Gd is $4f^7 5d^1 6s^2$ ants regarding sulphur is <i>inc</i> is mostly of S ₈ rings. asists of S ₂ molecules. ur is never less than +4 in it as	o increase in size of the central a 4) is: (3) $4f^7 5d^1 6s^2$ orrect? s compounds.	
Sol.: 80. Sol.: 81.	$X_{CH_3OH} = \frac{5.2}{5.2 + 55.55} = 0.086$ Correct choice: (2) Which of the following statemet (1) Nitrogen cannot form $d\pi$ -p π (2) Single N–N bond is weaker (3) N ₂ O ₄ has two resonance strue (4) The stability of hydrides income The stability of hydrides decreas Correct choice: (4) The outer electron configuration (1) 4f ⁸ 5d ⁰ 6s ² Outer electronic configuration of Correct choice: (3) Which of the following statemet (1) The vapour at 200°C consisting (2) At 600°C the gas mainly cont (3) The oxidation state of sulph (4) S ₂ molecule is paramagnetic	nt is <i>wrong</i> ? bond. than the single P–P bond. actures. reases from NH ₃ to BiH ₃ in ses from NH ₃ to BiH ₃ due t n of Gd (Atomic number: 64^{-1} (2) $4f^4 5d^4 6s^2$ of Gd is $4f^7 5d^1 6s^2$ ants regarding sulphur is <i>inc</i> is mostly of S ₈ rings. asists of S ₂ molecules. ur is never less than +4 in it as	o increase in size of the central a 4) is: (3) $4f^7 5d^1 6s^2$ orrect? s compounds.	

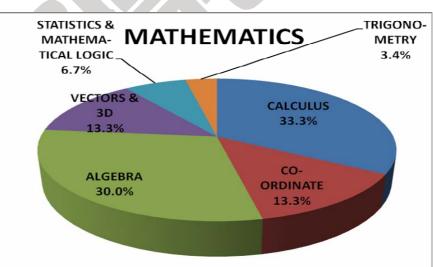
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AIEEE 2011 SOLUTIONS 20 *82. The structure of IF₇ is: (1) trigonal bipyramid (2) octahedral (3) pentagonal bipyramid (4) square pyramid The hybridisation of central atom is sp^3d^3 and its structure is pentagonal bipyramid. Sol.: **Correct choice: (3)** *83. Ozonolysis of an organic compound gives formaldehyde as one of the products. This confirms the presence of: (1) a vinyl group. (2) an isopropyl group. (3) an acetylenic triple bond. (4) two ethylenic double bonds. Sol.: Vinyl group on ozonolysis gives formaldehyde as one of the products. Correct choice: (1) *84. A gas absorbs a photon of 355 nm and emits at two wavelengths. If one of the emissions is at 680 nm, the other is at: (1) 325 nm (2) 743 nm (3) 518 nm (4) 1035 nm $\frac{1}{\lambda} = \frac{1}{\lambda_1} + \frac{1}{\lambda_2} \quad ; \quad \frac{1}{355} = \frac{1}{680} + \frac{1}{\lambda_2} \quad ; \quad \lambda_2 = 743 \text{ nm}.$ Sol.: Correct choice: (2) 85. Silver mirror test is given by which one of the following compounds? (3) Benzophenone (4) Acetaldehyde (1) Acetone (2) Formaldehyde Sol.: Correct choice: (4) 86. Which of the following reagents may be used to distinguish between phenol and benzoic acid? (1) Tollen's reagent (2) Molisch reagent (3) Neutral FeCl₃ (4) Aqueous NaOH Sol.: Phenol gives a violet colour with neutral ferric chloride solution. Correct choice: (3) 87. Phenol is heated with a solution of mixture of KBr and KBrO₃. The major product obtained in the above reaction is: (3) 2,4,6-Tribromophenol (1) 3-Bromophenol (2) 4-Bromophenol (4) 2-Bromophenol **Sol.:** $5Br^{-} + BrO_{3}^{-} + 3H_{2}O \Longrightarrow 3Br_{2} + 6OH^{-}$ Correct choice: (3) 88. In a face centred cubic lattice, atom A occupies the corner positions and atom B occupies the face centre positions. If one atom of B is missing from one of the face centred points, the formula of the compound is (1) AB₂ $(4) A_2 B$ (2) A_2B_2 $(3) A_2B_5$ Contribution by A atoms at corners = $8 \times \frac{1}{2} = 1$. Sol.: Contribution by B atoms at face centre = $5 \times \frac{1}{2}$ So, formula is $AB_{5/2}$ or A_2B_5 Correct choice: (3) *89. The entropy change involved in the isothermal reversible expansion of 2 moles of an ideal gas from a volume of 10 dm^3 to a volume of 100 dm³ at 27°C is: (2) 32.3 J mol⁻¹ K⁻¹ (3) 42.3 J mol⁻¹ K⁻¹ (4) 38.3 J mol⁻¹ K⁻¹ (1) $35.8 \text{ J mol}^{-1} \text{ K}^{-1}$ **Sol.:** ΔS for isothermal reversible expansion is given by $\Delta S = nR \ln \frac{V_2}{V_1} = 2 \times 8.314 \ln \frac{100}{10} = 2 \times 8.314 \times 2.303 \times 1 = 38.3 \text{ J mol}^{-1} \text{ K}^{-1}.$ Correct choice: (4) Identify the compound that exhibits tautomerism. *90. (1) Lactic acid (2) 2-Pentanone (3) Phenol (4) 2-Butene Sol.: $CH_3-C-CH_2-CH_2-CH_3 \Longrightarrow CH_3-C=CH-CH_2-CH_3$

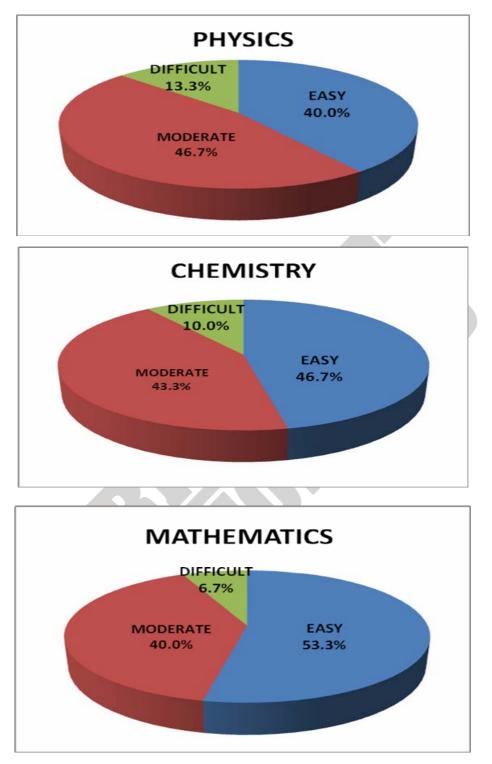
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Correct choice: (2)

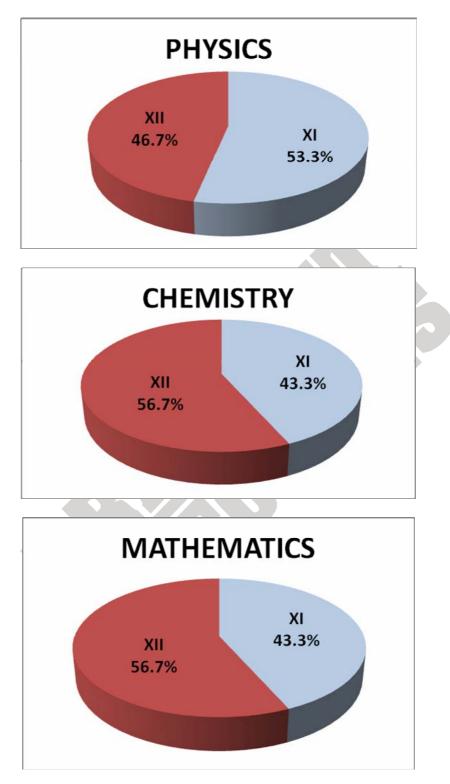




LEVEL OF DIFFICULTY



CLASSWISE



Read the following instructions carefully

- 1. The candidates should fill in the required particulars on the Test Booklet and Answer Sheet (side 1) with Blue/Black Ball Point Pen.
- 2. For writing / marking particulars on **side 2** of the Answer Sheet, use Blue/Black Ball Point Pen only.
- **3.** The candidate should not write their Roll Numbers anywhere else (except in the specific space) on the Test Booklet/Answer Sheet.
- **4.** Out of the four options given for each question only one option is the correct answer.
- **5.** For each incorrect response, one-fourth (1/4) of the total marks allotted to the question would be deducted from the total score. No deduction from the total score, however, will be made if no response is indicated for an item in the Answer Sheet.
- **6.** Handle the Test Booklet and Answer Sheet with care, as under no circumstance (except for discrepancy in Test Booklet Code and Answer Sheet Code) will another set be provided.
- 7. The candidates are not allowed to do any rough work or writing work on the Answer Sheet. All calculations/ writing work are to be done in the space provided for this purpose in the Test Booklet itself, marked 'Space for Rough Work'. This space is given at the bottom of each page and in 3 pages (Pages 21-23) at the end of the booklet.
- **8.** On completion of the test, the candidates must hand over the Answer Sheet to the Invigilator on duty in the Room/Hall. However, the candidates are allowed to take away this Test Booklet with them.
- **9.** Each candidate must show on demand his/her Admit Card to the Invigilator.
- **10.** No candidate, without special permission of the Superintendent or Invigilator, should leave his/ her seat.
- 11. The candidates should not leave the Examination Hall without handing over their Answer Sheet to the Invigilator on duty and sign the Attendance Sheet again. Cases where a candidate has not signed the Attendance Sheet a second time will be deemed not to have handed over the Answer Sheet and dealt with as an unfair means case. The candidates are also required to put their left hand THUMB impression in the space provided in the Attendance Sheet.
- **12.** Use of electronic/ Manual Calculator and any electronic Item like mobile phone, pager etc. is prohibited.
- **13.** The candidates are governed by all Rules and Regulations of the Board with regard to their conduct in the Examination Hall. All cases of unfair means will be dealt with as per Rules and Regulations of the Board.
- 14. No part of the Test Booklet and Answer Sheet shall be detached under any circumstances. Candidates are not allowed to carry any textual material, printed or written, bits of papers, pager, mobile phone, electronic device or any other material except the Admit Card inside the examination hall/ room.

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