

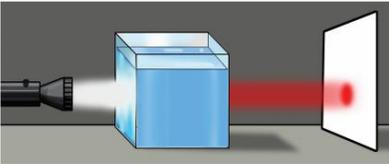
SSLC MODEL EXAMINATION 2017 FEBRUARY

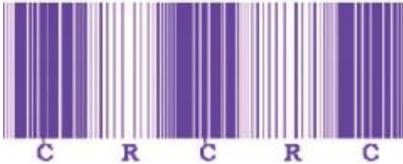
PHYSICS

Max.score:40

Time: 1 ½ hr

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| 1 | Radio waves | Score: 1 |
| 2 | (a) Nitrogen : Red (b) Frequency: 50 Hz | Score: ½ Score: ½ |
| 3 | <u>Advantages</u> <ul style="list-style-type: none"> • Hydrogen has more calorific value • It does not causes pollution <u>Disadvantages</u> <ul style="list-style-type: none"> • Explosive in nature • It cannot be stored easily, and difficult to transport. | Score: 1 Score: 1 |
| 4 | (a) <ul style="list-style-type: none"> • Bandages that heals wound easily • Manufacturing of batteries of increased efficiency • Producing paints, varnish • Creating display screen of low weights • Long lasting tennis ball (any four usage) (b) <ul style="list-style-type: none"> • When particles are converted to nano size the ratio between their surface area and volume increases comparatively, thereby improving their physical properties. | Score: 2 Score: 1 |

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| <p>5</p> | <p>(a) Voltage , $V= 240 \text{ V}$ Power $P= 40 \text{ W}$ Resistance , $R = \frac{V^2}{P} = \frac{240 \times 240}{40} = 1440 \Omega$</p> <p>(b)</p> <ul style="list-style-type: none"> • High melting point • High ductility • High resistivity • Ability to emit white light in the white hot condition | <p>Score: 2</p> <p>Score: 2</p> |
| <p>6</p> | <p>(A)</p> <p>(a) Different colors have different wavelength and different rate of scattering</p> <p>(b) Experiment with sodium thiosulphate and HCl</p>  <p>(c) When the size of particle is greater than the wave length of light , all the colors will scatter equally</p> <p>(B)</p> <p>(a) Scattering : It is the phenomenon of irregular or partial reflection of light is called scattering.</p> <p>(b) Tyndal effect: the scattering of light beam on colloidal solution (particle size less than 1 nm)</p> <p>(c)</p> <ul style="list-style-type: none"> • High wavelength so they can travel more distance • Scattering loss is minimum and can visible in night | <p>Score: 1</p> <p>Score: 2</p> <p>Score: 1</p> <p>Score: 1</p> <p>Score: 1</p> <p>Score: 1</p> |

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| | <p>(d) Dark , because in moon there is no atmosphere so there is no scattering of light occurs</p> | <p>Score: 1</p> |
| <p>7</p> | <p>(a) If the natural frequency of the body undergoing vibration is the same as that of the influencing body, the vibration is maximum and is called resonance</p> <p>(b) Resonance column</p> <p>(c) High pressures are experienced in places where the air particles are close. Such regions are called compression (C). Region of low pressure are rarefaction (R) , here air particle are not close.</p>  <p>Particles vibrate in a direction parallel to direction of propagation of wave.</p> | <p>Score: 1</p> <p>Score: 1</p> <p>Score: 2</p> |
| <p>8</p> | <p>(a) Sound coming out from the source gets reflected repeatedly from many parts of the building and reaches us. so many sound waves reaching our ears at the same time causes a boom sound called reverberation</p> <p>(b)</p> <ul style="list-style-type: none"> • Make the floor rough • Used folded curtains • Use carpet on floor • Construction of more windows and doors • Cushioning the seats • Avoid curved walls (any two methods) | <p>Score: 1</p> <p>Score: 1</p> |

| 9 | <p>(a) 0.6 A because in fig A bulbs are connected in series so current is same</p> <p>(b) Fig. B</p> <p>(c)</p> <ul style="list-style-type: none"> • Same voltage can obtain for each device • Can control each device individually | <p>Score: 1</p> <p>Score: 1</p> <p>Score: 2</p> | | | | | | | | | | | | |
|----------------|------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------|-------------------------------------------------|---|---|-------------|---------------------|---------|----------|-----------------------------------------------------------------------------------|-------|----------------|------------------|--------------|-----------------|
| 10 | <table border="1" style="margin-left: auto; margin-right: auto;"> <thead> <tr> <th style="text-align: center;">A</th> <th style="text-align: center;">B</th> <th style="text-align: center;">C</th> </tr> </thead> <tbody> <tr> <td style="text-align: center;">Green color</td> <td style="text-align: center;">Complementary color</td> <td style="text-align: center;">Magenta</td> </tr> <tr> <td style="text-align: center;">Inductor</td> <td style="text-align: center;"></td> <td style="text-align: center;">Henry</td> </tr> <tr> <td style="text-align: center;">Nuclear energy</td> <td style="text-align: center;">Non conventional</td> <td style="text-align: center;">Brown energy</td> </tr> </tbody> </table> | A | B | C | Green color | Complementary color | Magenta | Inductor |  | Henry | Nuclear energy | Non conventional | Brown energy | <p>Score: 3</p> |
| A | B | C | | | | | | | | | | | | |
| Green color | Complementary color | Magenta | | | | | | | | | | | | |
| Inductor |  | Henry | | | | | | | | | | | | |
| Nuclear energy | Non conventional | Brown energy | | | | | | | | | | | | |
| 11 | <p><u>Evaporation</u></p> <p>It is the process by which a liquid changes into vapour by absorbing heat from surroundings. This is a normal process that takes place in the surface of liquid at all temperature. it is a slow process</p> <p><u>Vapourisation</u></p> <p>Vapourisation is the process by which a liquid changes into its gaseous state at its boiling point.</p> <ul style="list-style-type: none"> • The temperature dependence of both are different | <p>Score: 2</p> | | | | | | | | | | | | |

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| <p>12</p> | <p>(a) No: of Primary turns, $N_P=20,000$ No: of secondary turns, $N_S = 30,000$ Voltage in the primary, $V_P= 160$ V Voltage in the secondary, $V_S = ?$ $\frac{V_S}{V_P} = \frac{N_S}{N_P} \therefore V_S = \frac{V_P \times N_S}{N_P} = \frac{160 \times 30000}{20000} = 240$ V</p> <p>(b) As no.of turn's increases in the secondary the flux linked with secondary increases thereby increases the induced emf in the secondary.</p> <p>(c) 500 W. Because power in the primary and secondary are equal.</p> | <p>Score: 2</p> <p>Score: 1</p> <p>Score: 1</p> |
| <p>13</p> | <p>(A)</p> <p>(a) Mass , $m = 5$ Kg Change in temperature , $\theta = 313\text{K} - 303\text{K} = 10$ K Quantity of heat , $Q = 209300$ J $Q = mc\theta$ $C = \frac{Q}{m\theta} = \frac{209300}{5 \times 10} = 4186 \text{ J/Kg K}$</p> <p>(b)</p> <ul style="list-style-type: none"> • The change in atmospheric temperature do not affect our body temperature quickly • Water is used as a coolant in radiators of engines • Land breeze and sea breeze <p>(B)</p> <p>(a) J / Kg</p> <p>(b) It is the quantity of heat absorbed by 1 Kg of solid to change into liquid state at its melting point without change in temperature</p> | <p>Score: 2</p> <p>Score: 2</p> <p>Score: 1</p> <p>Score: 1</p> |

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| | (c) <ul style="list-style-type: none"> Glaciers do not melt as a whole at the same time Ice creams does not melt fast | Score: 2 |
| 14 | c. 55000 Kj / Kg | Score: 1 |

| Chapter no | Chapter name | No of questions (optional) | Marks | % weight |
|------------|-------------------------------------|-----------------------------|-----------|-------------|
| 1 | Wave motion | 5 | 6 | 15% |
| 2 | Effects of electric current | 3 | 5 | 12.5% |
| 3 | Electromagnetic induction | 2 | 4 | 10% |
| 4 | Power transmission and distribution | 4 | 5 | 12.5% |
| 5 | Heat | 3(3) | 6 | 15% |
| 6 | Colours of light | 5(4) | 6 | 15% |
| 7 | Electronics and modern technology | 3 | 4 | 10% |
| 8 | Energy management | 3 | 4 | 10% |
| | | Total | 40 | 100% |

❖ NB: % weight of score for each chapter may vary