#### SCIENCE AND TECHNOLOGY (Theory) Delhi Compartment— 2006

#### **General Instructions:**

- 1. The question paper comprises two Sections, A and B. You are to attempt both the Sections.
- 2. The candidates are advised to attempt all the questions of Section A separately and Section B separately.
- 3. All questions are compulsory.
- 4. There is no overall choice. However, internal choice has been provided in some questions. You are to attempt only one option in such questions.
- 5. Marks allocated to every question are indicated against it.
- 6. Question numbers 1—4 in Section A and 17— 18 in Section B are very short answer questions and are of 1 mark each. These are to be answered in one word or one sentence each.
- 7. Question numbers 5—8 in Section A and 19— 20 in Section B are short answer questions and are of 2 marks each. These are to be answered in 30—40 words each.
- 8. Question numbers 9—14 in Section A and 21—23 in Section B are also short answer questions and are of 3 marks each. These are to be answered in 40—50 words each.
- 9. Question numbers 15—16 in Section A and 24 in Section B are long answer questions and are of 5 marks each. These are to be answered in 70 words each.

## **SECTION – A**

<ul><li>Q. 1. Give an example of an endothermic reaction.</li><li>Q. 2. What kind of lens is used in the spectacles of a person suffering from manual spectacles.</li></ul>	<b>(1)</b> nyopia (near-	
sightedness)?	(1)	
<b>Q. 3.</b> Name a metal which is both ductile as well as malleable.	(1)	
<b>Q. 4.</b> There are two electric bulbs,		
i. marked 60 W; 220 V and		
ii. marked 100 W; 220 V. Which one of the two has a higher resistance?	(1)	
<b>Q. 5.</b> What is meant by the term "magnetic field lines"? List two properties of magnetic field lines. <b>(2)</b>		
Or		
With the help of a neat diagram describe how you can generate induced current in a circ	cuit.	
<b>Q. 6.</b> Write two observations you would make when quick lime is added to water.	(2)	
<b>Q.</b> 7. Choose a metal out of the following which reacts with hot water but not with cold water:		
Sodium, Magnesium, Iron		
Mention the products formed during the reaction.	(2)	
<b>0. 8.</b> Out of the two, equatorial and polar orbits of man-made satellites, which one i	s suitable for	

- Q. 8. Out of the two, equatorial and polar orbits of man-made satellites, which one is suitable for collection of data for weather prediction? Why?(2)
- Q. 9.

- a. What is meant by pH of a solution?
- b. State one difference between a strong electrolyte and a weak electrolyte. Give one example of each. (3)
- Q. 10. In a household 5 tubelights of 40 W each are used for 5 hours and an electric press of 500 W for 4 hours every day. Calculate the total electrical energy consumed by the tubelights and press in a month of 30 days.
- **Q. 11.** What is an alloy? Name the constituents of 22- carat gold. Why is 24-carat gold converted to 22-carat gold ? **Or**

Draw a labelled diagram to show the extraction of sulphur by Frasch process.

# Q. 12.

- a. Name the device used to convert
  - i. solar energy into heat, and
  - ii. solar energy into electricity.
- b. Explain the principle of working of a wind-mill.

## Q. 13.

- a. Name the functional group present in propanone (acetone).
- b. What is the product formed when prdpane is reduced? Name the reducing agent used...

(3)

(3)

- c. What happens when propanone is oxidised by alkaline KMnO<sub>4</sub>?
- Q. 14. What are Jovian planets? Why are they so called? Write any two special features of Jovian planets.(3)

## Q. 15.

- a. Name an important ore of iron. Write its formula.
- b. How is this ore concentrated?
- c. Describe with chemical equations, the reactions taking place in the furnace to obtain iron from the concentrated ore. (5)

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Give reasons for the following;

- a. Metals conduct electricity.
- b. Metals generally do not form compounds with hydrogen.
- c. A piece of zinc placed in blue copper sulphate solution decolourises it.
- d. Alumina is dissolved in molten cryolite for electrolysis to obtain aluminium metal.
- e. Nitrogen gas is used, to preserve food.

## Q. 16.

- a. State the relation between object distance, image distance and focal length of a spherical mirror.
- b. Draw a ray diagram to show the image formed by a concave mirror when an object is placed between pole and focus of the mirror.
- c. A concave mirror of focal length 15 cm forms an image of an object kept at a distance of 10 cm from the mirror. Find the position, nature and size of the image formed by it. (5)

# **SECTION B**

<b>Q. 17.</b> What is a gene? (1	l)
Q. 18. What is vegetative propagation? (1	l)
Q. 19. Mention two harmful effects of each of the following pollutants that are emitted from	n motor
vehicles:	
i. Carbon monoxide	
ii. Lead (2	2)
Or	
Define eutrophication. State its two harmful effects.	
<b>Q. 20.</b> What is sustainable 'development? Suggest any one method to achieve it. (2	2)
<b>Q. 21.</b> What is pollination? Name its two types. How do they differ from each other? (3)	<b>3</b> )
Q. 22. Draw a diagram of human alimentary canal showing duodenum, small intestine, liv	ver and
pancreas. (3	3)
Or	
State the role of the following in the human respiratory system:	
<b>Q. 23.</b> What is blood transfusion? What could be the consequence if proper matching of bloo	od is not
done during transfusion? Give the reason of such a consequence. (3	8)
Q. 24.	
a. What is a reflex action? Give its two examples. Illustrate the pathway followed by a n	nessage
from the receptor in a reflex arc.	

b. Name the sympathetic and para sympathetic systems of eye. (5)