

Quarterly Common Examination Sep – 2023 Science – Answer Key

	IX Standard	1	
Question No	Answer Key		Marks
1	(b) 10 quintals		
2	(d) acceleration		1
3	(a) density		1
4	(b) Juice		1
5	(c) Homogeneous Mixture		1
6	(b) an isotope		1
7	(c) Laws of octaves		1
8	(b) Coelentera e		1
9	(c) Crocodile		1
10	(b) Sclerenchyma		1
11	(c) thigmotropism		1
12	(a) photosynthesis		1
13	• Pitch of the screw gauge is the distance between wo successive screw threads. It is measured by the ratio of dist nce travelled on the pitch scale to the number of rotations of the herd scale.		2
	S.No Distance The setuel length of the path	Displacement	1
14	travelled by a moving body irrespective of the direction	moving body in a particular direction	•
	Scalar quantity	Vector quantity	1
15	a) less b) atmospheric pressure		1 1
16	 When a magnet is br ught near he mixture containing saw dust naphthalene and ron filings, it attracts the iron filings. Thus iron filings are separated The mixture of workthalene and some dust are put in a shine disk and 		1
	• The mixtue of haphthalene and saw dust are put in a clima dish and cove ed wi a perforated asbestos sheet. An inverted funn 1 is placed over the bestos sheet.		1
17	K - 2, L = 8, M - 18		1
	28 electrons – Nickel		1
18	gills, lungs, skin and pharyngeal region.		2
19	a. Sclereids Sclerenchyma b. Chloroplast Chlorenchyma c. Companion cell Phloem d. Tracheids Xylem		$4 imes rac{1}{2} = 2$
20	• During meiosis, a germ cell or gar cells. As a result of fertilization tw egg or zygote. Therefore only if gar take place.	mete divides to make four new sex yo gamates join together to form an metes are produced, fertilization can	2

21	• Blood platelets play a major role in the clotting of blood whenever there is a wound/injury. If blood platelets are removed from the blood, clotting of blood will not occur. In case of any injury/surgery etc., blood will be lost from the body in excess and may even prove to be fatal.		2	
	Density of Mercury = 13600	kg m ⁻³		
	Density of water at 4°C = 100 Density density - Den	00 kg m ⁻³ sity of mercury		Ι
22	Densi	ty of water at 4°C		
	$=\frac{13600}{1000}$	$\frac{\text{kg m}^{-3}}{1}$		1
	Relative Density = 13.6	kg m		1
	Mass	Weig	t	
	1. It is a fundamental quantity.	It is a derived quantity.		
	2. It has magnitude alone – scala	It has magnitude and	direction – vector	
	quantity.	quantity.		
22	3. It is the amount of matte	er It is the normal force ex	erted by the surface	Any 4
23	contained in a body.	on the object against g	ravitational pull.	$(4 \times 1 = 4)$
	4 Remains the same everywhere	Var es from place to pl		
	5 It is measured using physic:	It is measured using sn	ring halance	
	balance			
	6. Its unit is kilogram	Its unit is newton		
		its unit is newton		
			. 1 . 1 . 1	
	• when an object moves wi	th constant speed along	a circular path, the	2
24	motion is called uniform ci	rcular motion.		
	b) Example :			2
	• The earth moves around the sun in the uniform circular motion.		2	
	The moon moves in unifor	m circular motion arou	d the earth.	
	a) False			1
25	A compound can be broken	n into simpler ubst nces	chemically.	1
25	b) Certain solids change directly to a gas witho t pa sing through the liquid is		2	
	called sublimation.		2	
	• Elements with large difference i properties were included in the same			
	group. Eg: Hard metals like copper (Cu) and silver (Ag) were included			
	along with soft metals like sodium (Na) and potassium (K).			
	along with solt motals into sourain (14) and potassiani (11).			
	• No proper position could be given to the element hydrogen Non-			
	metallic hydrogen w s placed along with metals like lithium (Li)		_	
26	sodium (Na) and notassium (K)		4	
	• The increasing der of atomic mass was not strictly followed			
	• The increasing der of atomic mass was not surctly followed throughou			
	$F_{\alpha} C_{\alpha} N T_{\alpha} \& I$			
	No lac for isotopes in the periodic table			
	• No fac for isotopes in the	e periodic table.		
	a) Lithin s dium and potassium atoms have only one electron in their valence shell. Hence their			
	coefficient is 1	ave only one electron in their v	aience snen. Hence inen	2
	Elements	Electron Configuration	Valancy	Z
	1 Lithium .	2,1	1	
	2 sodium 2	2, 8 , 1	1	
	3 potassium	288,1	1	
27	<i>b</i>)			
	Beryllium, magnesium and calc um atoms	have two electrons in their val	ence	
	shell, so their valence is 2.		T7 7	
	<i>Elements</i>	Electron Configuration	Valancy	
	1 Beryllium 2 2 magnasium	2,2	2	
	2 magnesium 3 calcium	2,0,2	2	
	5 cuicium	2,0,0,2	ے ا	

			2	
	Aquatic habits of amphibians:			
	• The larva of amphibians (tadpole) lives in water and breathes with gills.		2	
	• External fertilization occurs in frog with water as a medium of		Z	
	fertilization.	6		
20	• The adult frog has webbed feet	to swim in water.		
28	• The skin is moist and glandular	which helps in Respiration.		
	Terrestrial habits of amphibians:	I I I I I I I I I I I I I I I I I I I		
	• The adul s live on land and breathe through the lungs. Bucco-pharvnx		2	
	lso helps in Respiration.		2	
	• The forelimbs are short and help	p to hop on land.		
	• Complex tissues are made of more than	one type of cells that work together as a unit.		
	• Types : Two types		1	
	1. Xylem			
	2. Phloem			
	1) Xylem		1	
	• xylem tracheids			
29	• xylem fibres			
-	• xyl m vessels		1	
	• xylem parenchyma.			
	2) Phloem			
	• Sieve elements,			
	Companion cells			
	Phloem fibres		1	
	Phloem			
	Phototronism	Photonasty		
30	Movement of a plant part towards the light.	Mo em nt of a plant part is a response to		
50		light	4	
	Eg: Shoot of a plant.	Eg Moonflower, Taraxacum Officinale.		
	Stow at al transpiration	I aution law transmination		
	Loss of water from plants throu h stomata	Lenicular transpiration		
31	Loss of water from plants infour h stomatu.	the lenticels	4	
51	90-95% of transpiration in a plant takes place	A very small percentage of water is lost by	r -	
	through stomata only.	through plants lenticular transpiration.		
	W = 98 N	- ^ ^		
	$g = 9.8 \text{ m/s}^2$		_	
	$\widetilde{W} = mg$		2	
32	$m = \frac{W}{M}$			
	g 98		2	
	$m = \frac{1}{9.8}$			
	The mass of an object $m = 10 \text{ kg}$			
			l (Figure)	
			(1 (3410)	
33				
	Š u D A			
	N D N D	A		
	Se n D	A		
	v D 0 Time −	A t E		

	Figure shows the change in velocity with time for an uniformly accelerated		
	object. The object starts from the point D in the graph with velocity, u. Its velocity keeps increasing and after time, t it reaches the point B on the graph.		
	(i) First equation of motion:		
	u = OD = EA, $v = OC = EB$, $t = OE = DA$	2	
	\approx From the graph we know that $AB = DC$		
	By definition		
	Acceleration = Change in velocity / Time		
	= (Final velocity – Initial velocity)/Time		
	at = (OC-OD) / OE	2	
	= DC / OE = DC / t		
	DC = at = AB		
	From the graph, $EB = EA + AB$		
	v = u + at (1)	2	
	(ii) Second equation of motion:		
	From the graph the distance covered by the object during time, t is given by		
	the area of quadrangle		
	DOEB $S = Area$ of the quadrangle DOEB		
	= Area of the rectangle DOEA + Area of the triangl DAB		
	$= (AE \times OE) + \frac{1}{2} \times (AB \times DA)$		
	$S = ut + \frac{1}{2} at^2 \dots (2)$		
	(iii) Third equation of motion :		
	We see that the distance covered by the object during time t given by the		
	area of the quadrangle DOEB. Here, DOEB is a trapezi m. Then,		
	S = Area of trapezium DOEB		
	$= \frac{1}{2} \times \text{Sum of length of parallel side} \times \text{Distance between parallel sides}$		
	$= -\frac{1}{2} \times (\mathbf{U} + \mathbf{D}) \times \mathbf{U} = (\mathbf{Since}_{\mathbf{a}} - (\mathbf{v} - \mathbf{u})/t) \cdot \mathbf{t} = (\mathbf{v} - \mathbf{u})/2$		
	$S = \frac{1}{2} \times (u + v) \times t$ (Since, $u = (v - u) / t$), $t = (v - u) / a$		
	$2as = v^2 - u^2$		
	$v^2 = u^2 + 2as$		
	Mercury Barometer		
	Vacuum		
	Pressure exerted		
	by the coultrm of mercury		
	Atmospheric 760 mm	2	
	pressue	$3\frac{1}{2}$	
	Surface of mercury		
	1 It is designed by Three regionali		
OR	Construction :		
	2. Mercury Bar meter consists of long glass tube closed at one end and opened at the other.		
	3. Mercury f d through open end and close that end by thumb and open it after immersing it into		
	a trough of mercury.		
	Working: A The Barometer works by balancing the marcury in the glass tube against the outside air pressure		
	5. If air pressure increases, it pushes more of the mercury un into the tub	$\frac{1}{3-}$	
	6. If air pressure decreases, more mercury drains from the tub.	2	
	7. As vacuum cannot exert pres u e Mercury in the tube provides a precise measure of air pressu e		
	which is called atmospheric pressure.		
	8. It is used i alb ratory or weather station.		

	Atom of Oxygen ¹⁶ ₈ 0	Atom of Sulphur ³² ₁₆ s	
34	BP Bn Electronic Config. 2, 6 Valency = 2	Electronic Config. 2, 8, 6 Valency = 2	7
OR	 Brownian Movement: Brownian movement is a kinetic property. When colloidal solutions are viewed under a powerful microscope, it can be seen that colloidal particles are moving constantly and rapidly in zig-zag directions. The Brownian movement of particles is due to the unbalanced bomba dment of the particles by the molecules of the dispersion medium. 		$3\frac{1}{2}$
	 Tyndall (1869) obse solution, the path of a This phenomenon is cone. This phenomenon is a 	rved that when a strong beam of light f c sed on a colloidal the beam becomes visible. known as Tyndall effect and the il uminated path is called Tyndall not observed in case of true olu n. Laser beam Pure water Colloidal solution light beam not visible) (light beam visible)	$3\frac{1}{2}$
35	 Arthropoda is the largest The organisms h The body is se m The exosk le The co om c cat The d ot have system. Th insects shed Small Arthropod species breathet th Land Arthropod tracheae. Excretion occurs in crabs and prav Insects spiders, scorpions are sorted 	t phy um. ve jointed legs n nted into head, thorax and abdomen. is made up of chitin. vity is filled with haemolymph (blood). e defined blood vessels. This is called open c rculatory the exoskeleton and this process is called moulting. ls absorb oxygen through the body and larger aquatic hrough book gills. s breathe through a system of tiny body tubes called through malphigian tubules and through green glands vns crabs, shrimps, butterflies, millipedes, centipedes, and ne arthropods.	7
OR	Permanent tissues:		

• Permanent tissues are those in which, growth has stopped either completely or for the time being. At times, they become meristematic partially or wholly.	
Different types of simple normanent tissue :	
Different types of simple permanent tissue :	l
• Simple tissue: Simple tissue is homogeneous-composed of structurally and functionally simila cells. Eg : Parenchyma, collenchyma, and sclerenchyma	l
scierencinyma	l
Parenchyma Aerenchyma	7
Parenchyma	l
• Parenchyma are simple permanent tissue composed of livi cel	l
 Parenchyma cells are thin-walled oval rounded or polygon l in shape 	l
with well-developed spaces among them	l
• In aquatic plants, parenchyma possesses intercellular air, paces and is	1
named as aerenchyma	l
• When exposed to light parenchyma cells may dev lop chloroplasts and	l
are known as chlorenchyma	l
Functions:	l
• Parenchyma may store water in many succu ent and xerophytic plan s.	l
• It also serves the functions of storage of food reserves, absorption.	l
buovancy, secretion, etc	1
Collenchyma:	1
• Collenchyma is a living ti su found beneath the epidermis.	1
• Cells are elongated wit unevenly thickened non-lignified walls. Cells	l
have rectangular oblique or tapering ends and persistent protoplast.	l
• They possess thick primary non-lignified walls.	1
Functions:	1
They provide mechanical support for growing organs Sclerenchymà:	l
• Scierenchyma consists of thick-walled cells which are often lignified.	l
• Scierenchyma cells do not possess living protoplasts at maturity.	l
Scierenchyma cells are grouped into	l
• fibers nd (ii) sclereids.	1

